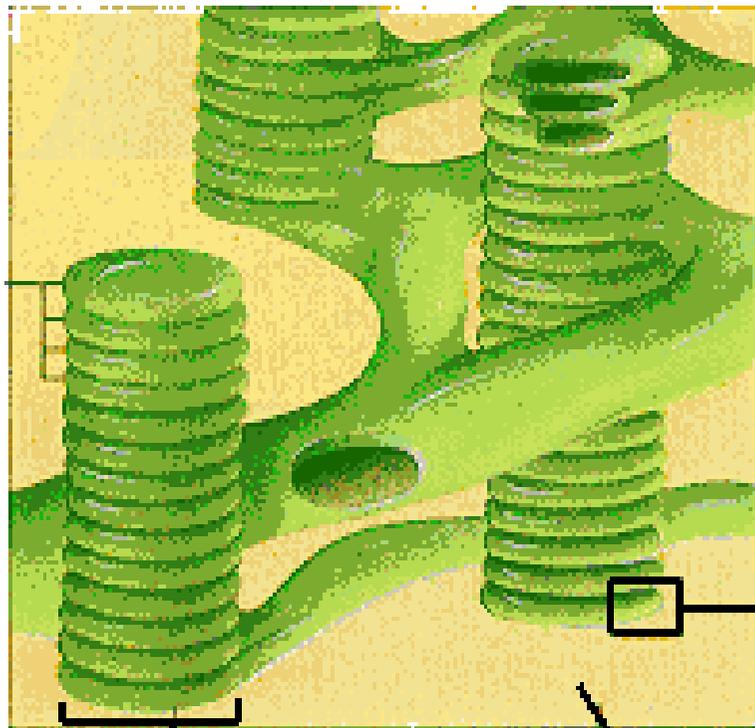
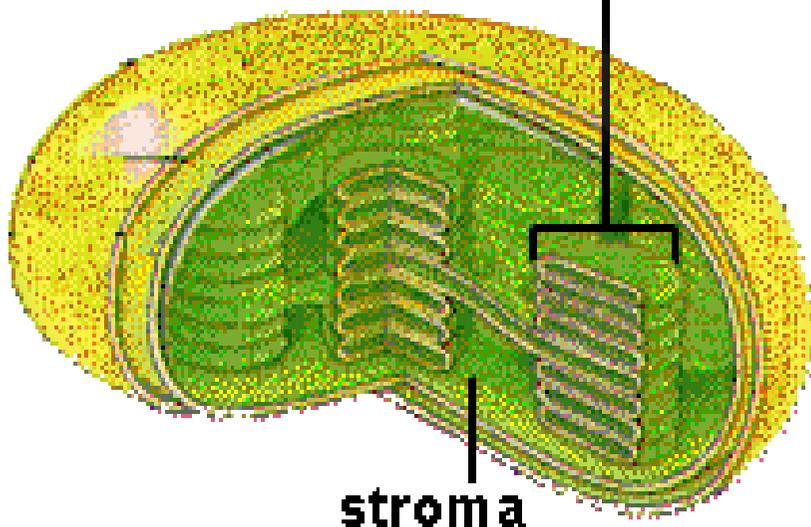


Introduction to Biophysics of Photosynthesis

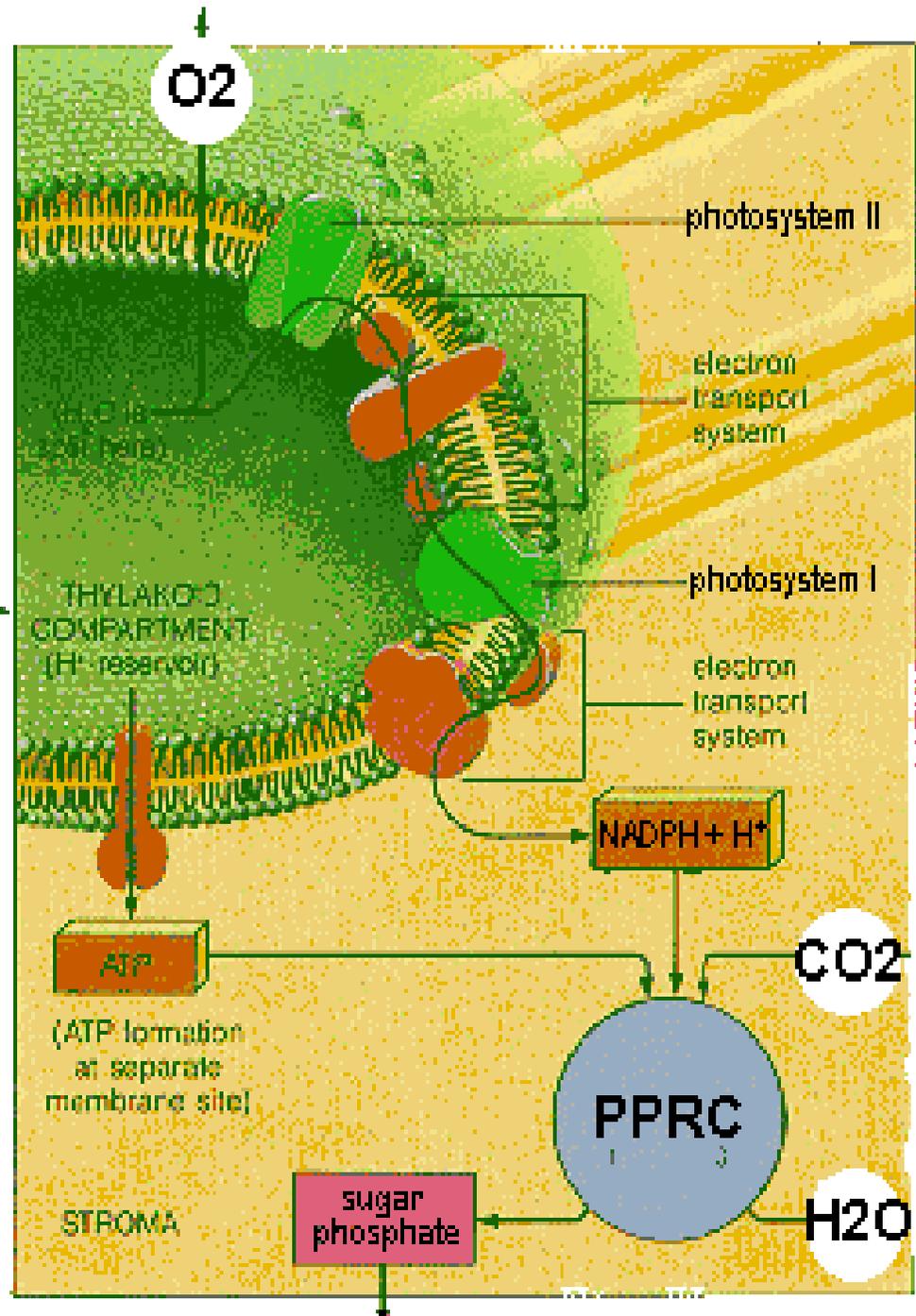


granum

stroma



stroma



O₂

photosystem II

electron transport system

photosystem I

electron transport system

THYLAKOID COMPARTMENT (H⁺ reservoir)

NADPH + H⁺

ATP

(ATP formation at separate membrane site)

CO₂

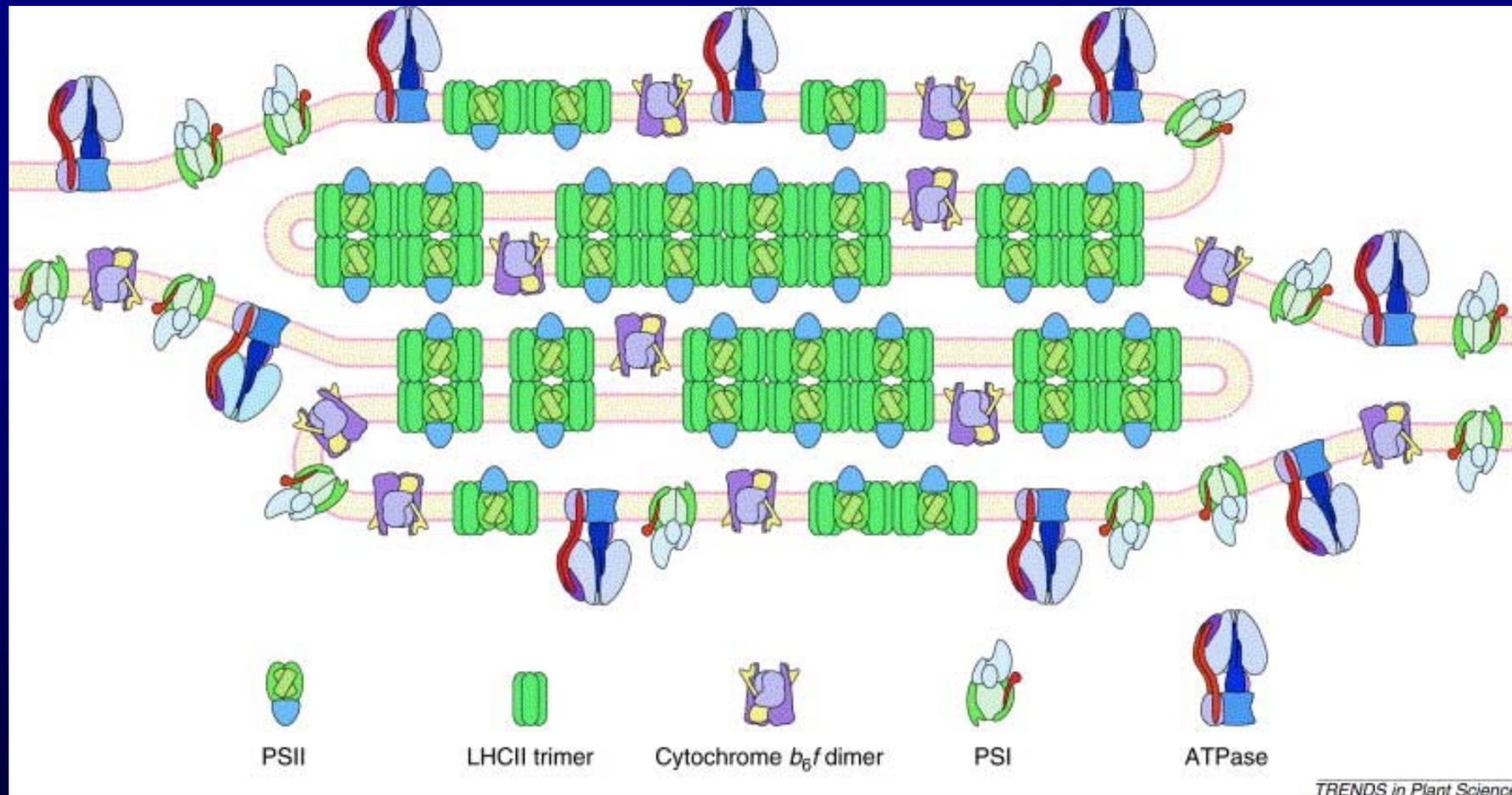
PPRC

H₂O

sugar phosphate

STROMA

Influence of steric hindrance on grana stacking

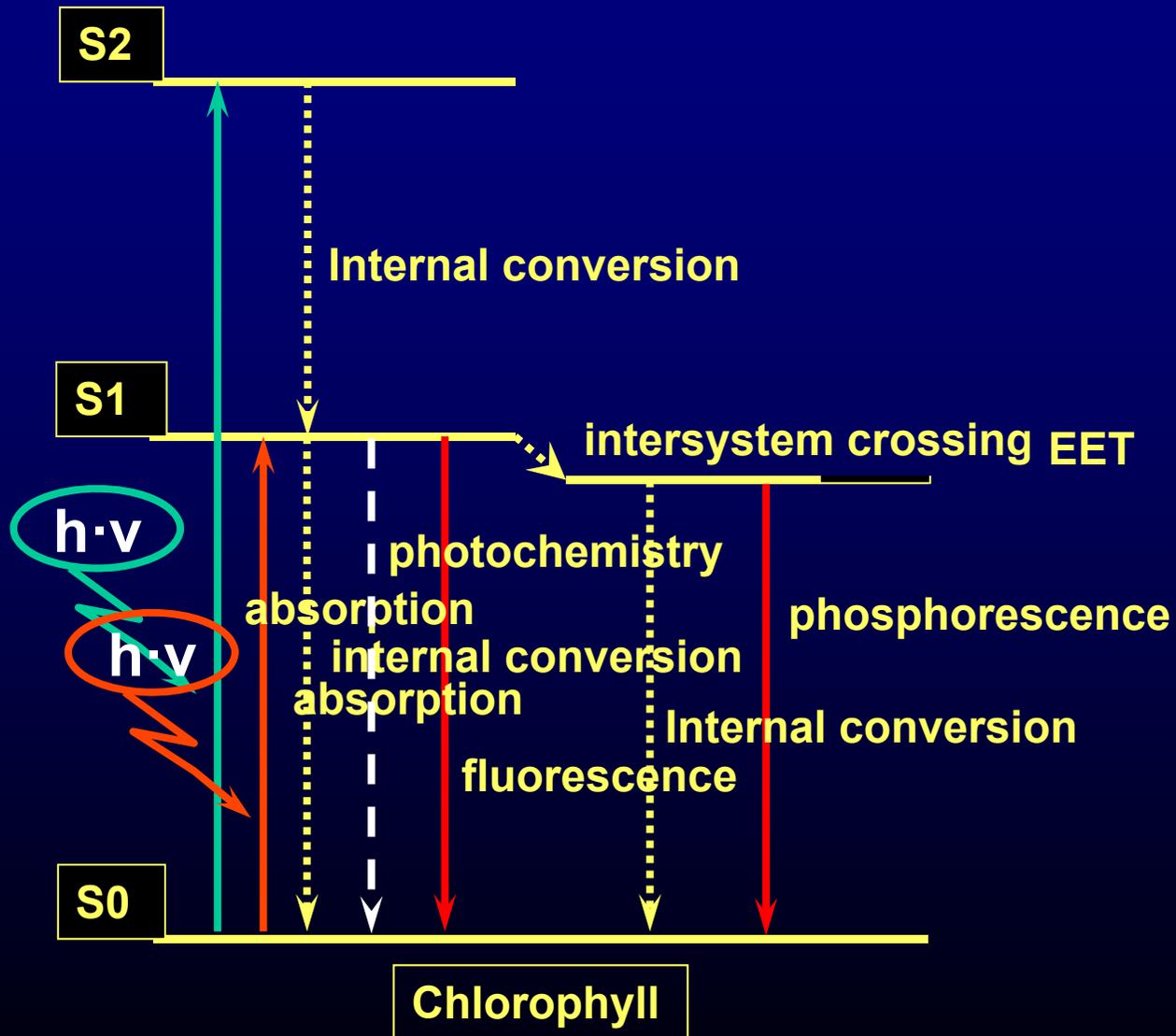


From: Allen JF, Forsberg J (2001) TIBS 6, 317–326

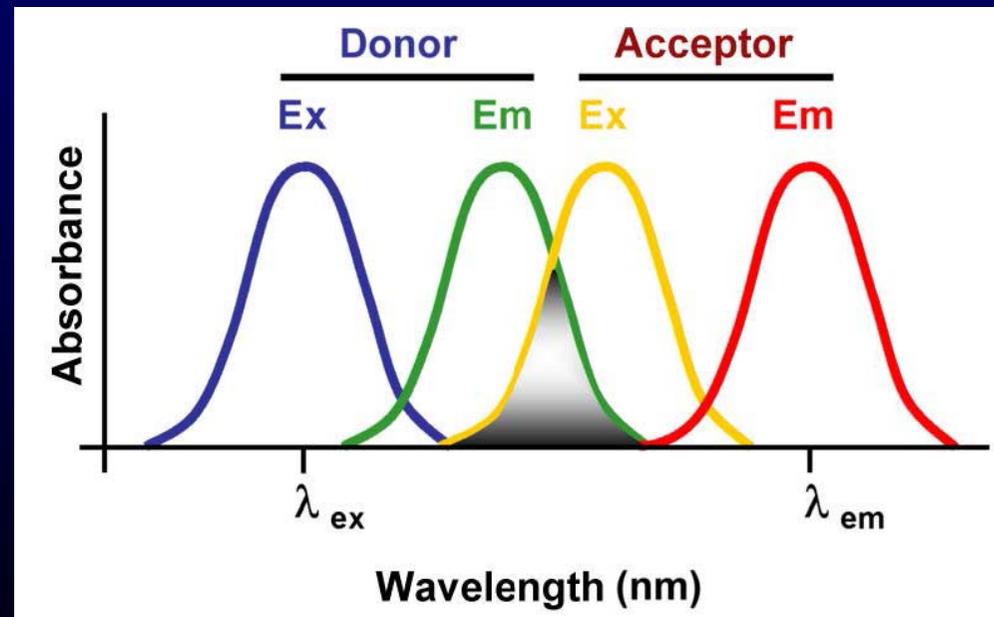
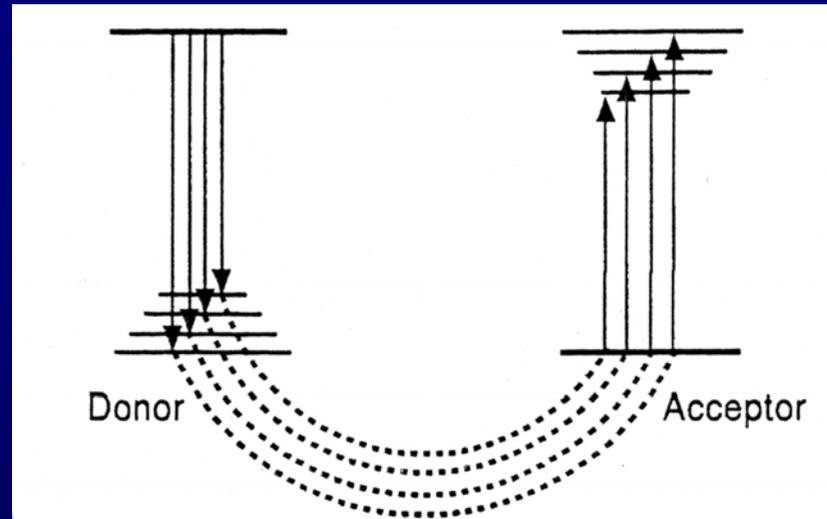
Mechanism of grana stacking

- 1. LHCII tends to aggregate
- 2. thylakoids containing a lot of LHCII will stick together, forming grana. PSII RC nicely fits in because it does not protrude much out of the membrane
- 3. The more bulky PS I RC and the most bulky ATPase go into stroma regions

Necessary for energy transfer: stable S1-state



Necessary for energy transfer: Overlap of emission/absorption bands

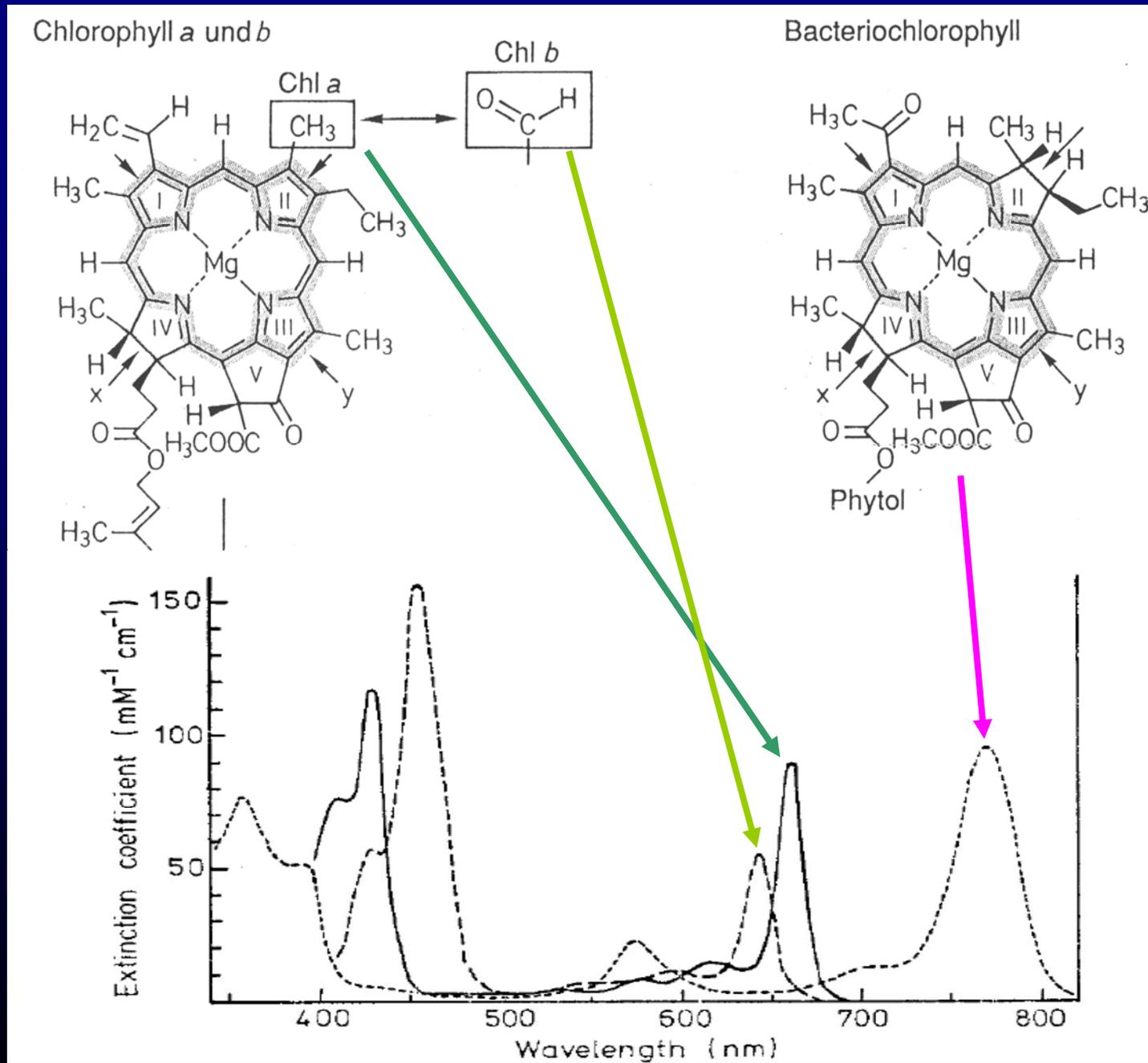


From: bio.libretexts.org

Adjustment of absorption bands by chemical modification

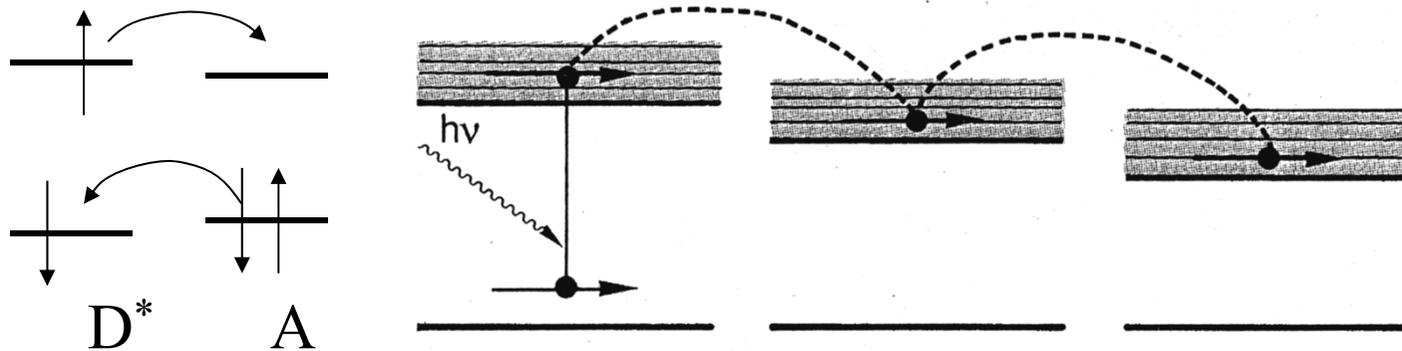
From: Lawlor
DW (1990)
Thieme,
Stuttgart,
377S

From: Barber J
(1978) Rep
Prog Phys 41,
1158-99

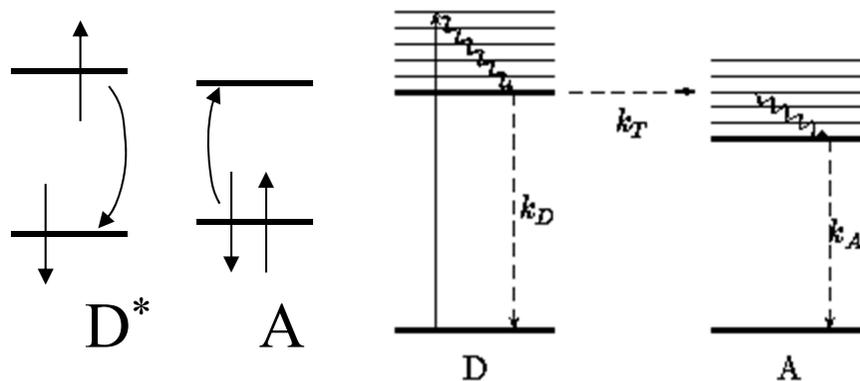


Mechanisms of energy transfer between chlorophylls

Short distance, requires overlap of molecular orbitals (\rightarrow only Chls in extremely short distance to each other, e.g. special pair) : direct transfer of S1 excited state (Dexter-Mechanism)



Larger distance, requires overlap of absorption/emission spectra: Transfer by inductive Resonance („Förster-Mechanism“)

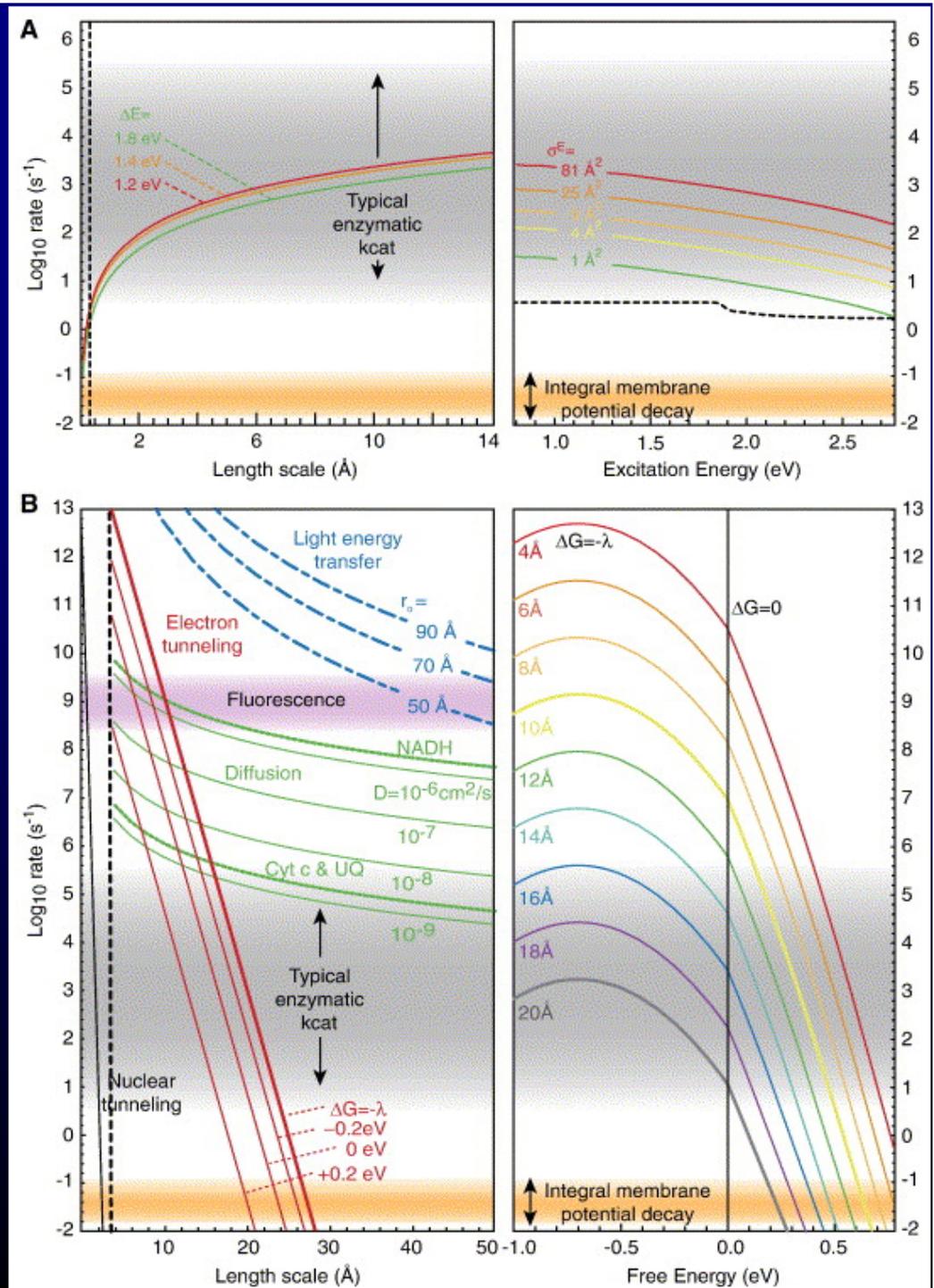


$$\Gamma_{DA} = k_D \left(\frac{R_0}{R} \right)^6$$

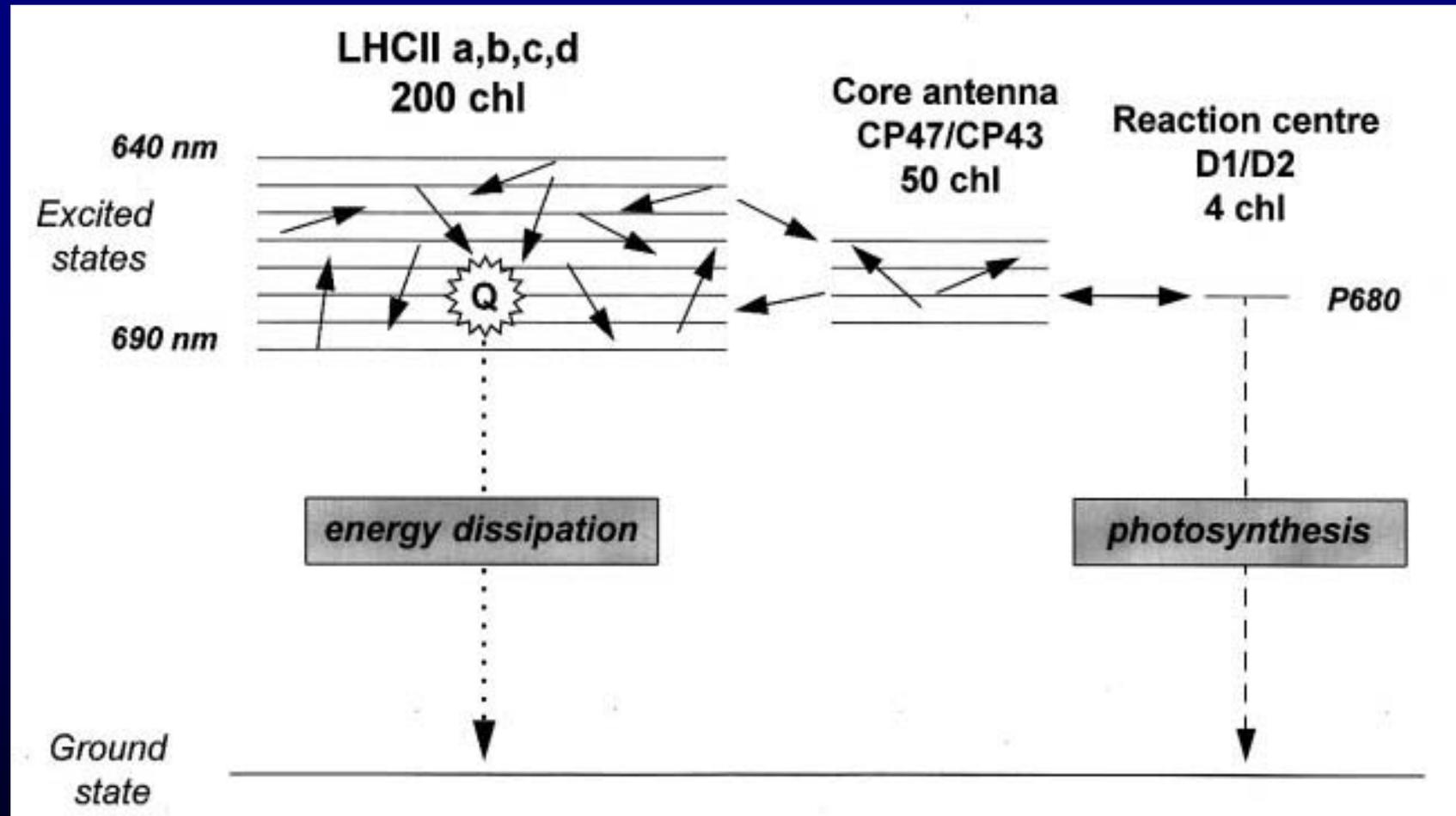
$$R_0^6 = 8.8 \times 10^{17} \frac{\kappa^2}{n^4} J$$

Comparison of other Energy transfer mechanisms

- For all processes, speed of energy transfer decreases with increasing distance.
- This limits the rate and efficiency of enzymatic and non-enzymatic processes. The longer the transfer time, the higher the risk of energy loss by unwanted processes
- Light energy transfer is fast and covers large distances, but required re-absorption and thus is not very efficient
- Electron tunnelling is fast for very short distances, but very slow for longer distances → most relevant <10Å.
- Diffusion speed decreases less with increasing distances, therefore it becomes faster than tunnelling at more than 10-20Å.

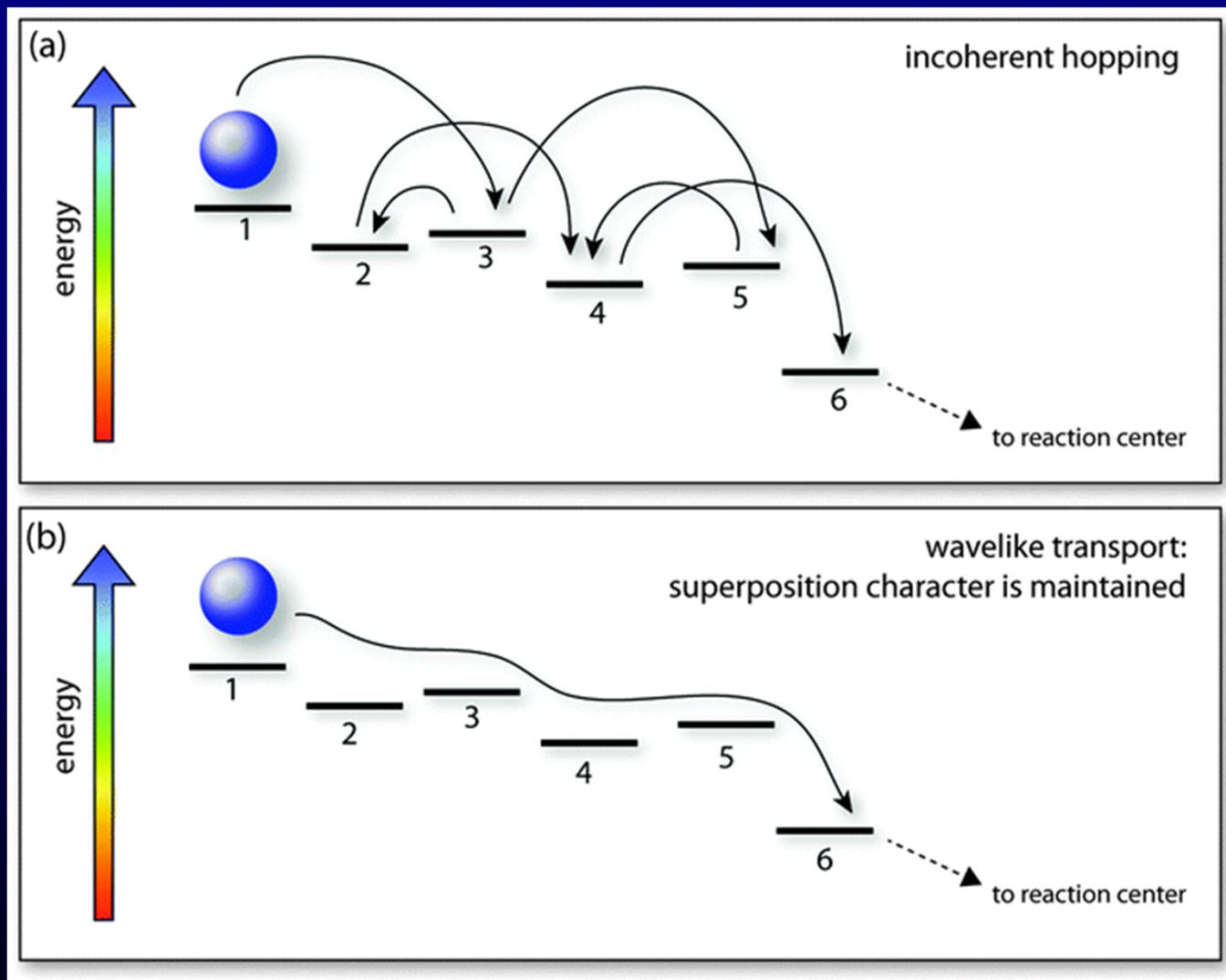


Energy transfer – funnel principle (II): Scheme in higher plants



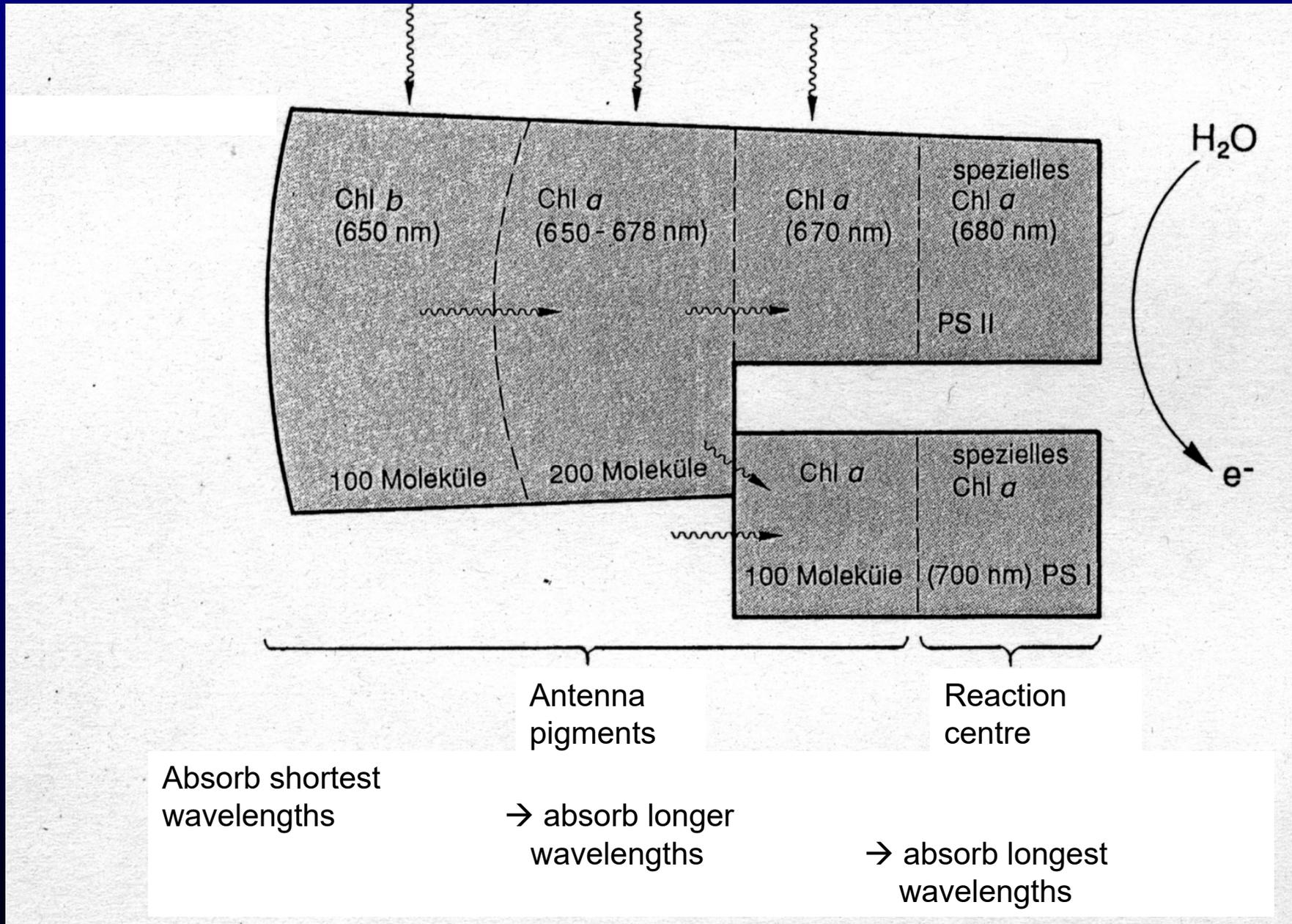
From: Horton P, Ruban AV, Walters RG (1996) Annu Rev Plant Physiol Plant Mol Biol 47: 655-84

Energy transfer – funnel principle (II): debated modern view

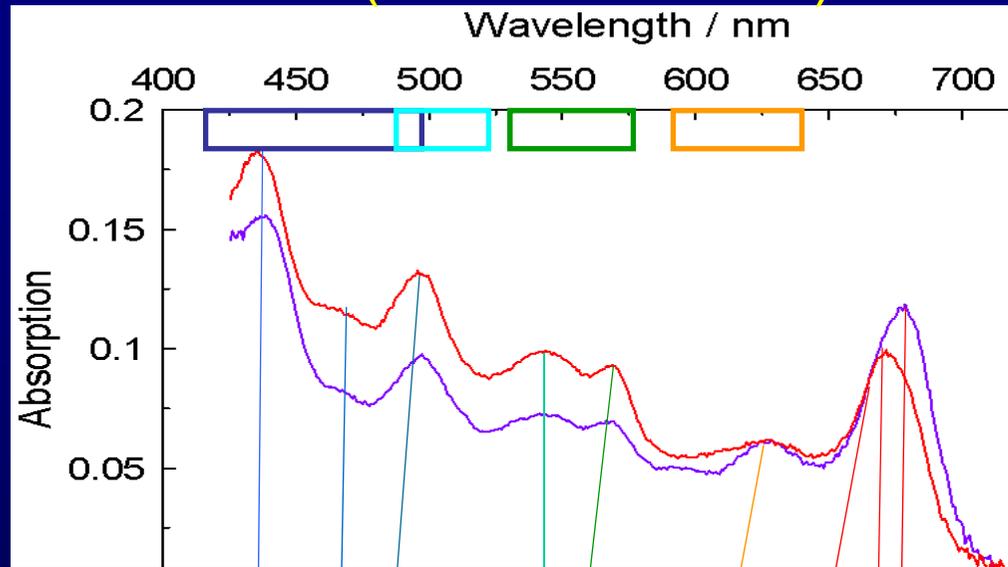


From: Collini E (2013) Spectroscopic signatures of quantum-coherent energy transfer. *Chemical Society Reviews* 42, 4932-4947

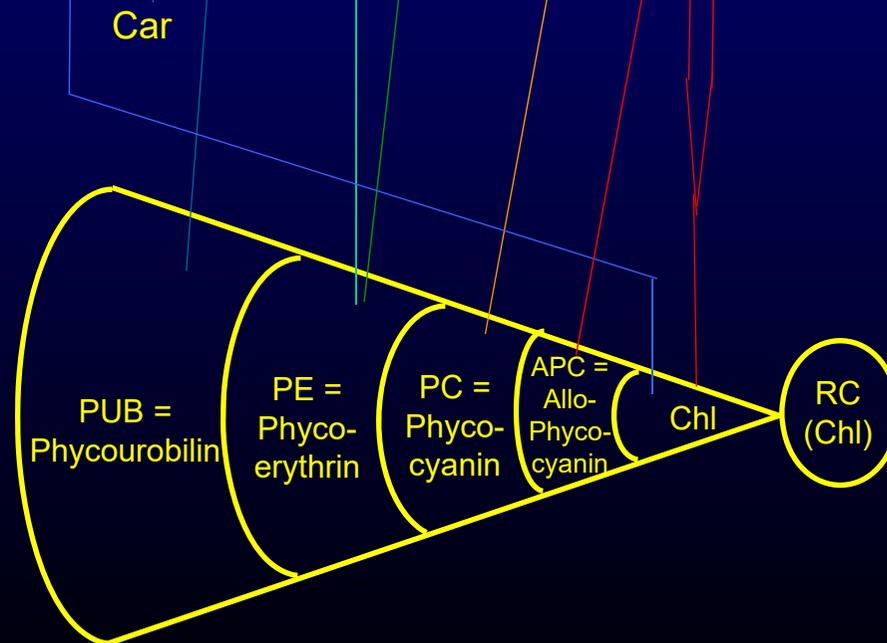
Energy transfer – funnel principle (II): Scheme in higher plants



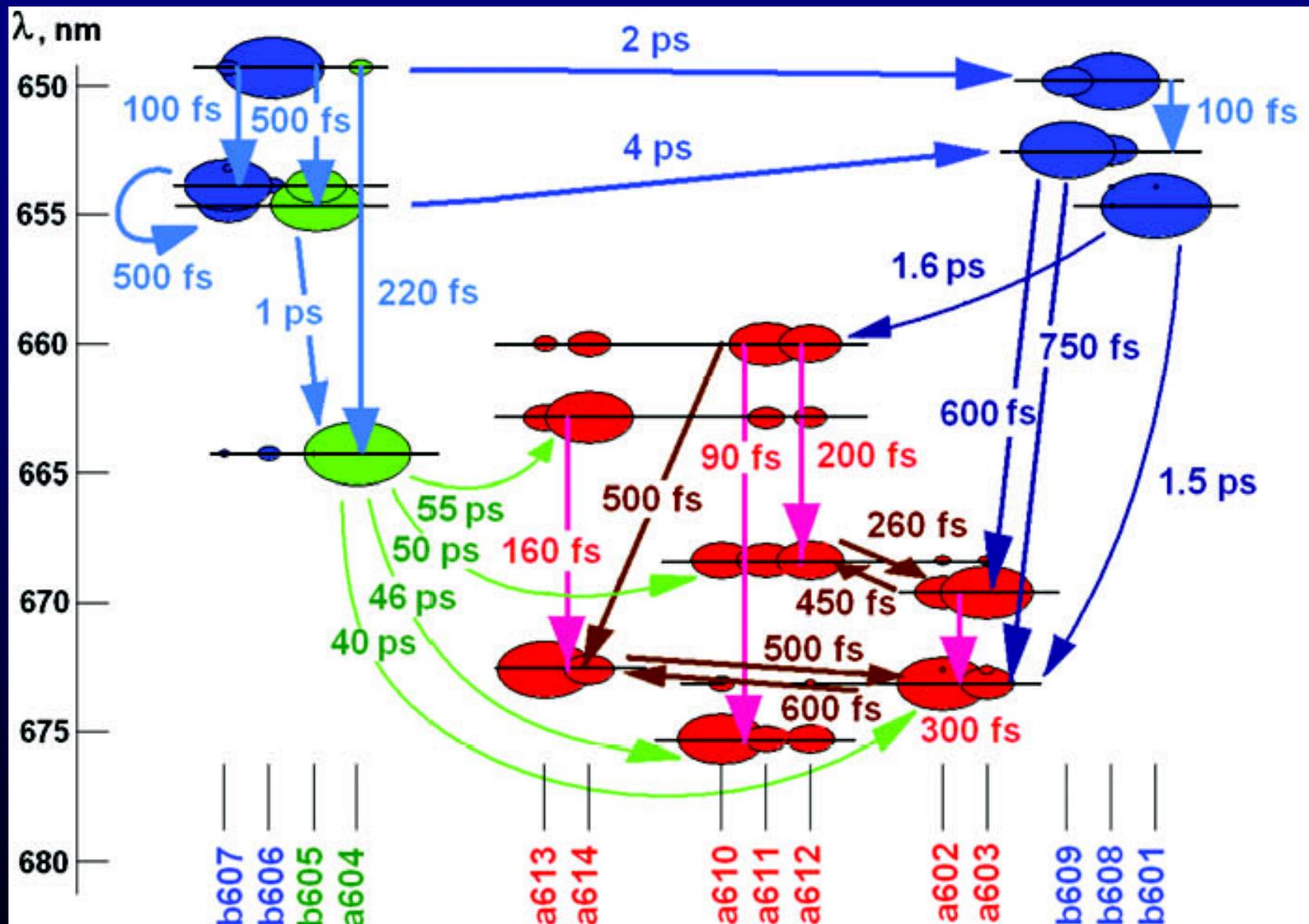
Energy transfer – funnel principle (II): Scheme in cyanobacteria (*Trichodesmium*)



Transmission of filters for selective excitation

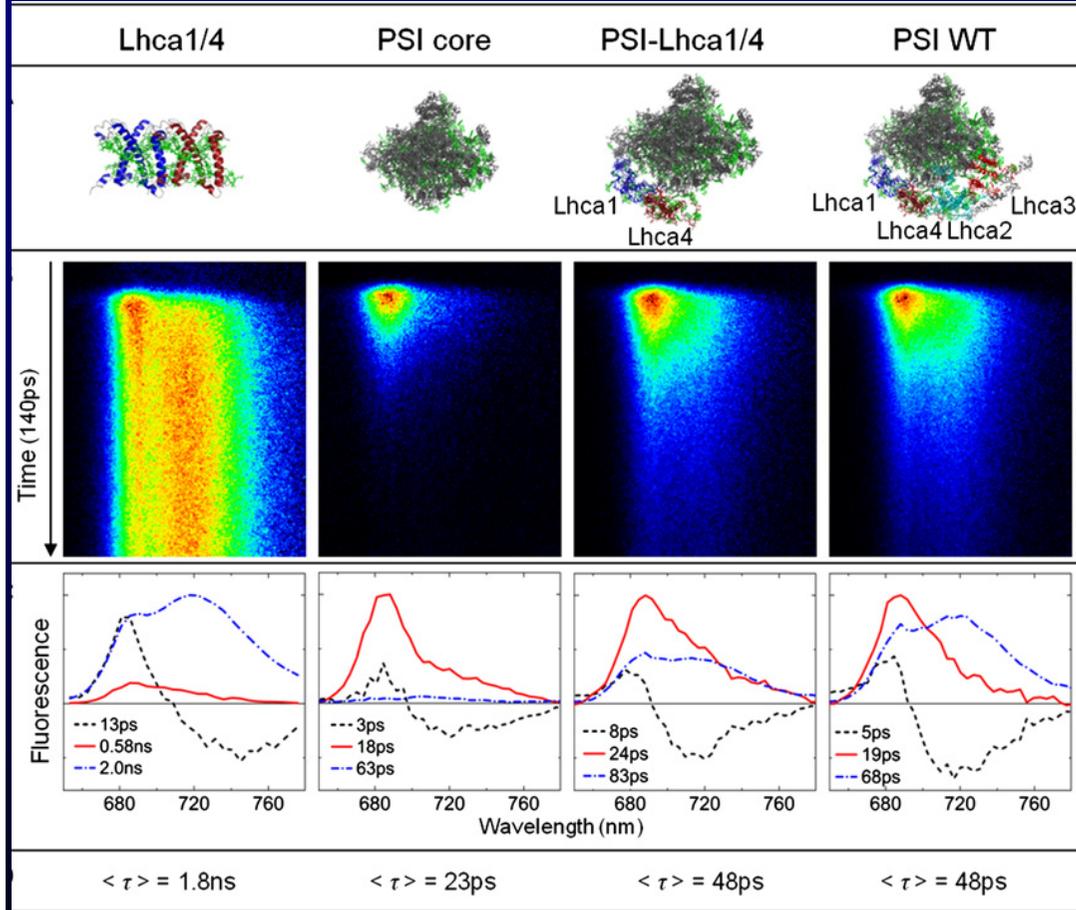


Energy transfer – funnel principle (III): Transfer times between Chls towards & in PSIIRC

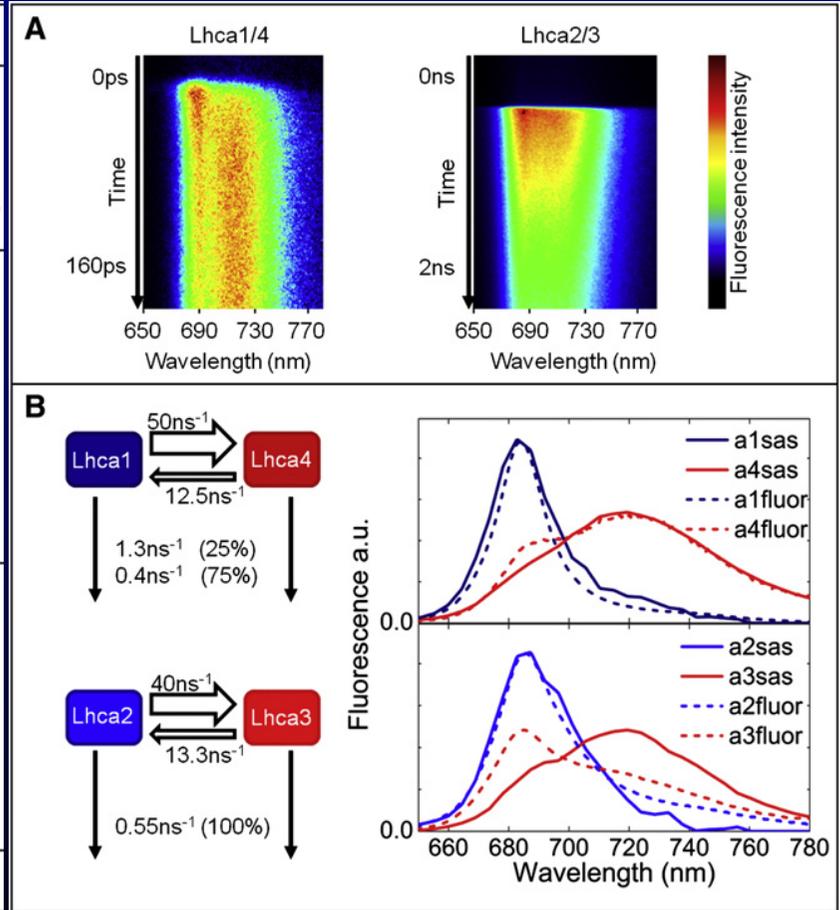


Photosynthesis related Proteins with metal centres

1. Excitation transfer times between light harvesting complexes



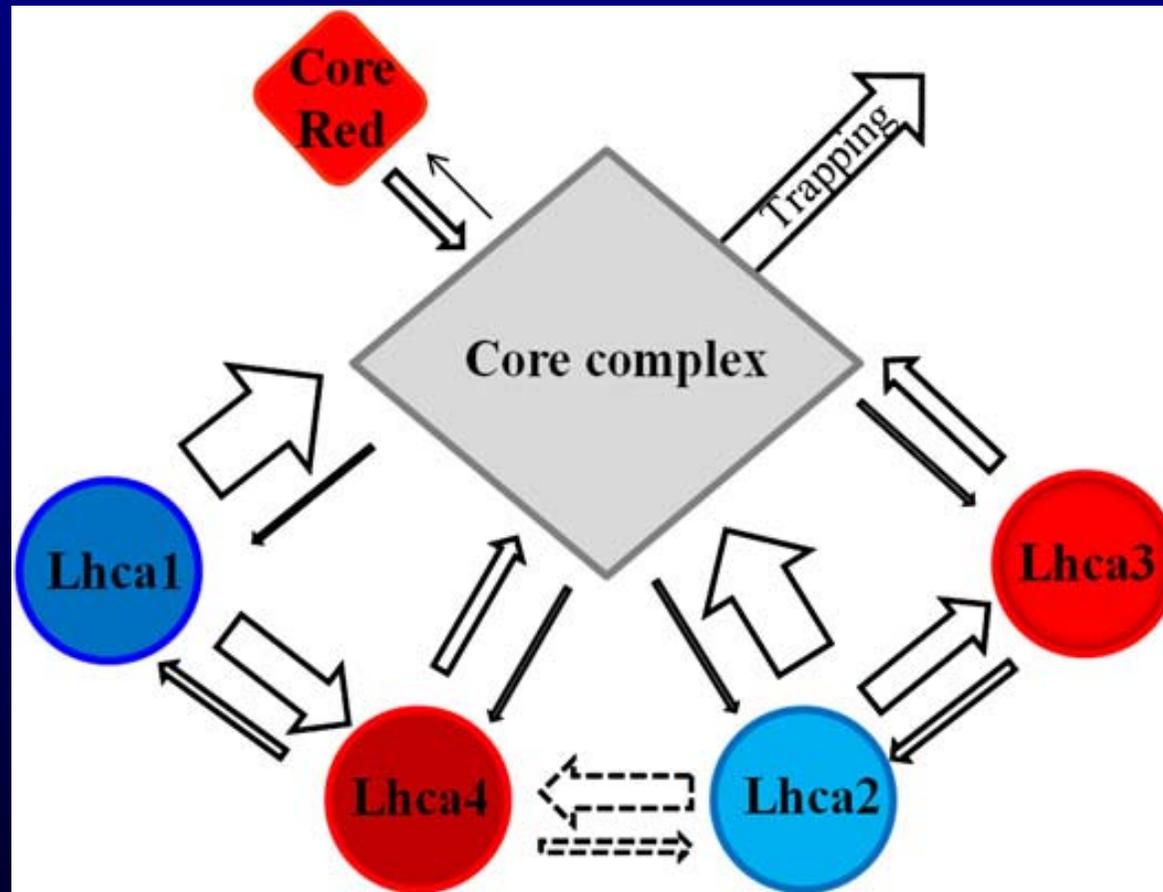
From: Wientjes E_et al (2011) BiophysJ101, 745-54



From: Wientjes E_et al (2011) BiophysJ 100, 1372-80

Photosynthesis related Proteins with metal centres

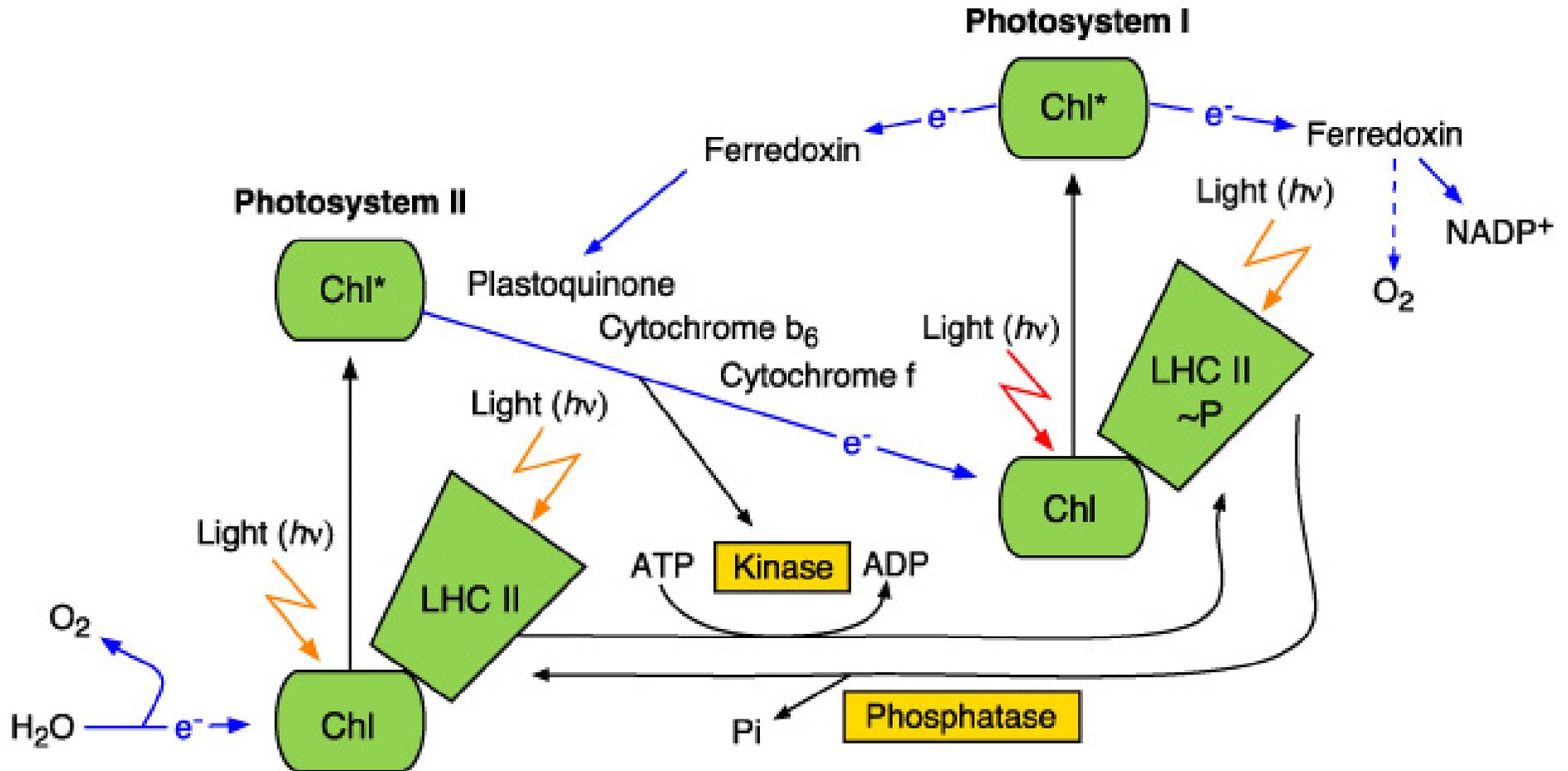
1. Excitation tranfer times between light harvesting complexes



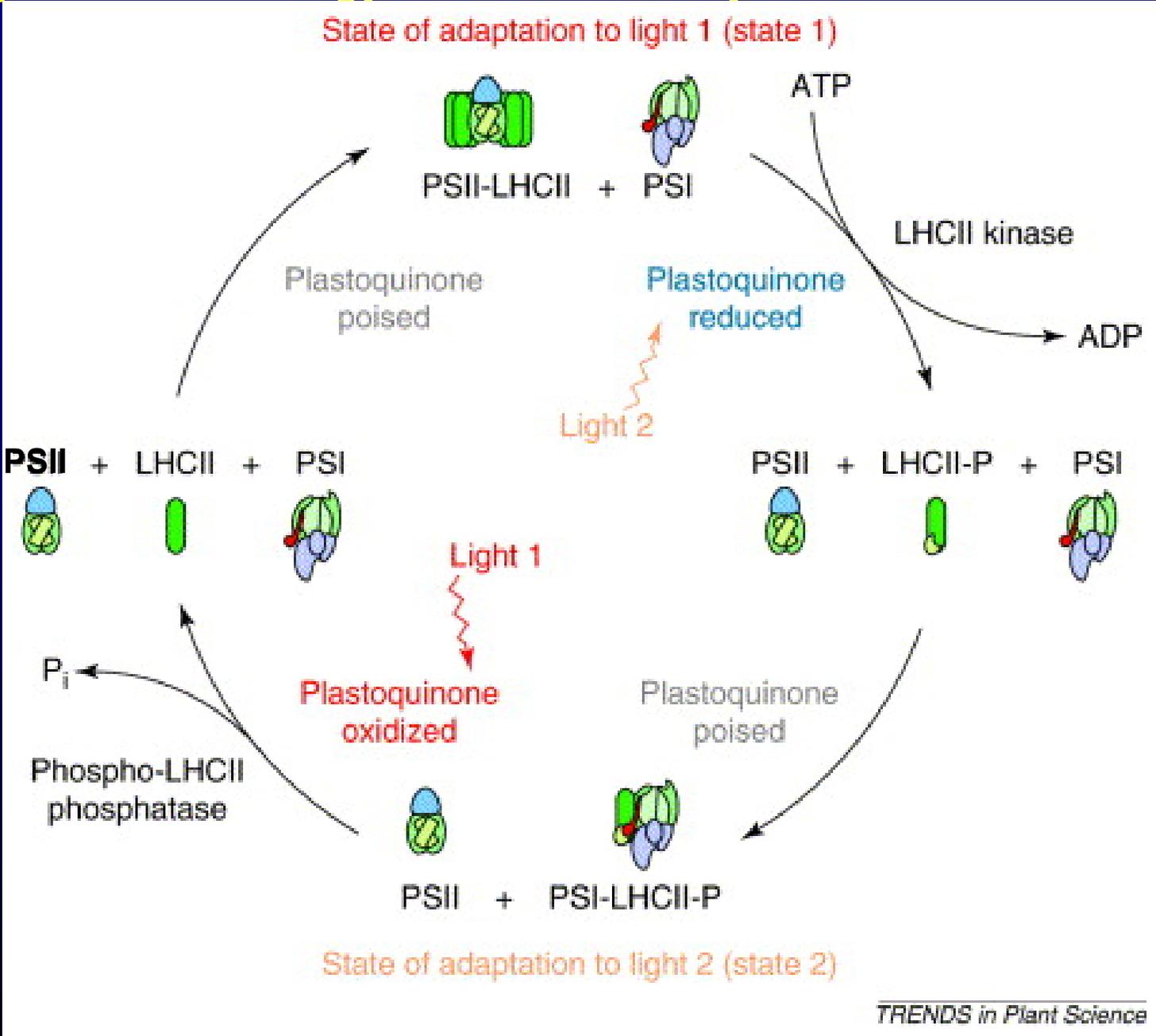
From: Wientjes E_ et al (2011) BiophysJ101, 745-54

Regulation of energy transfer (I): the principle of „state transitions“

Higher plants, many algae



Regulation of energy transfer: The cycle of state transitions

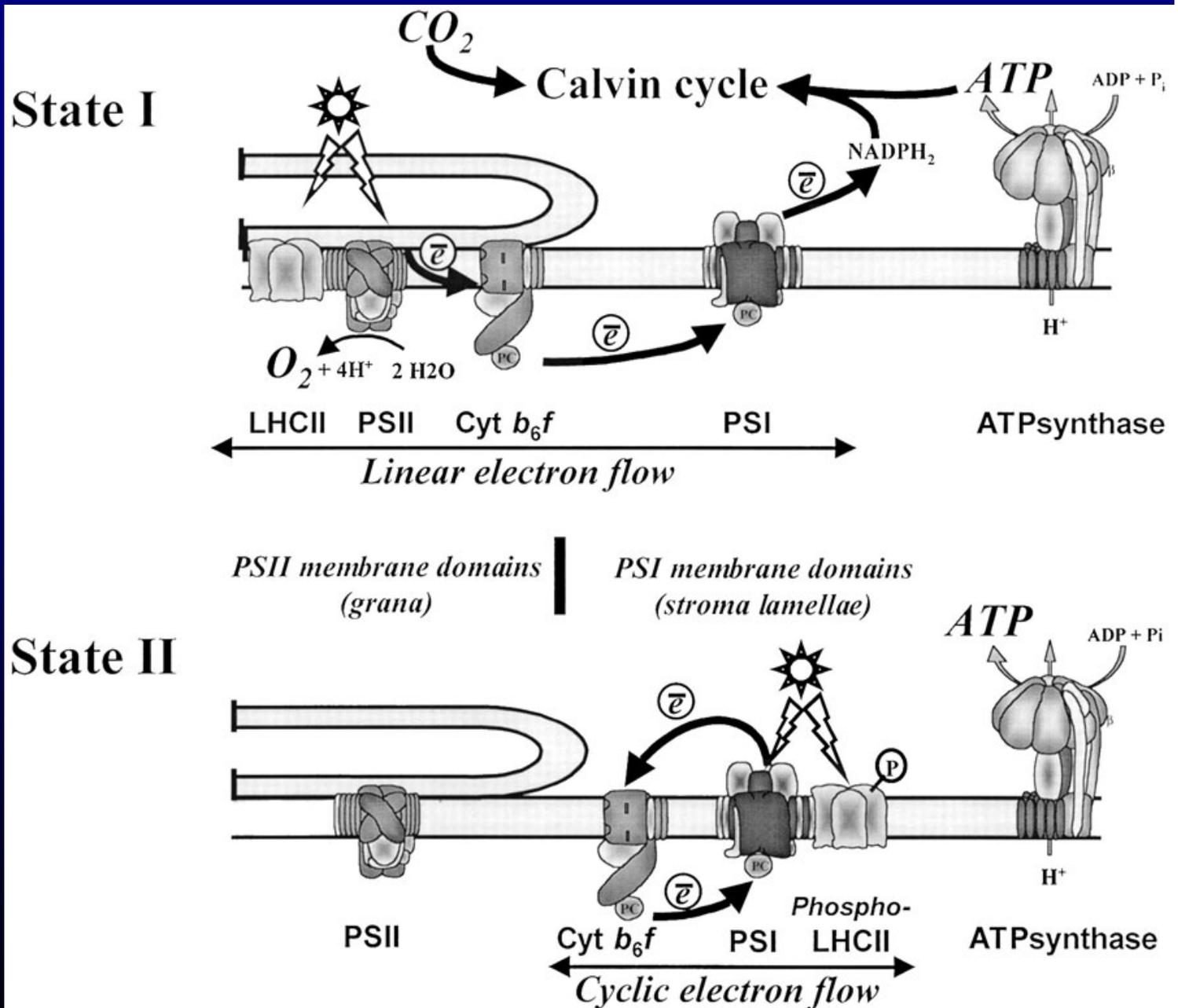


TRENDS in Plant Science

From: Allen JF, Forsberg J (2001) TIBS 6, 317–326

Regulation of energy transfer: another view of „state transitions“

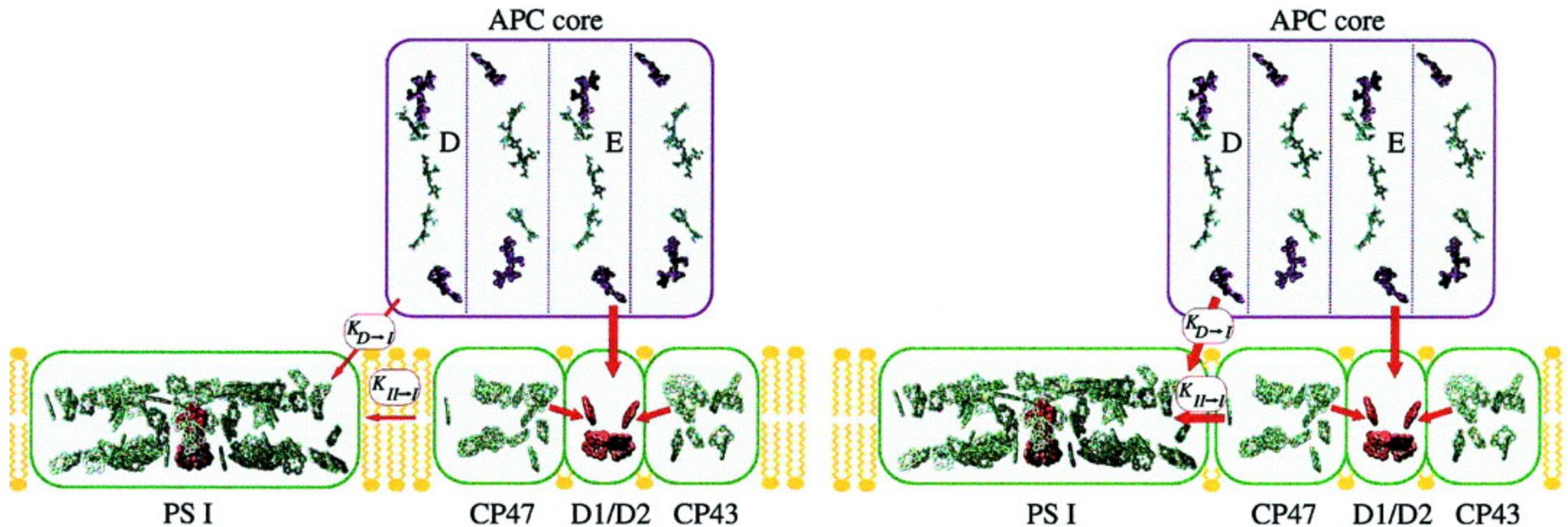
Alternative view of the function of state transitions



Regulation of energy transfer (I): „state transitions“ in cyanobacteria and red algae

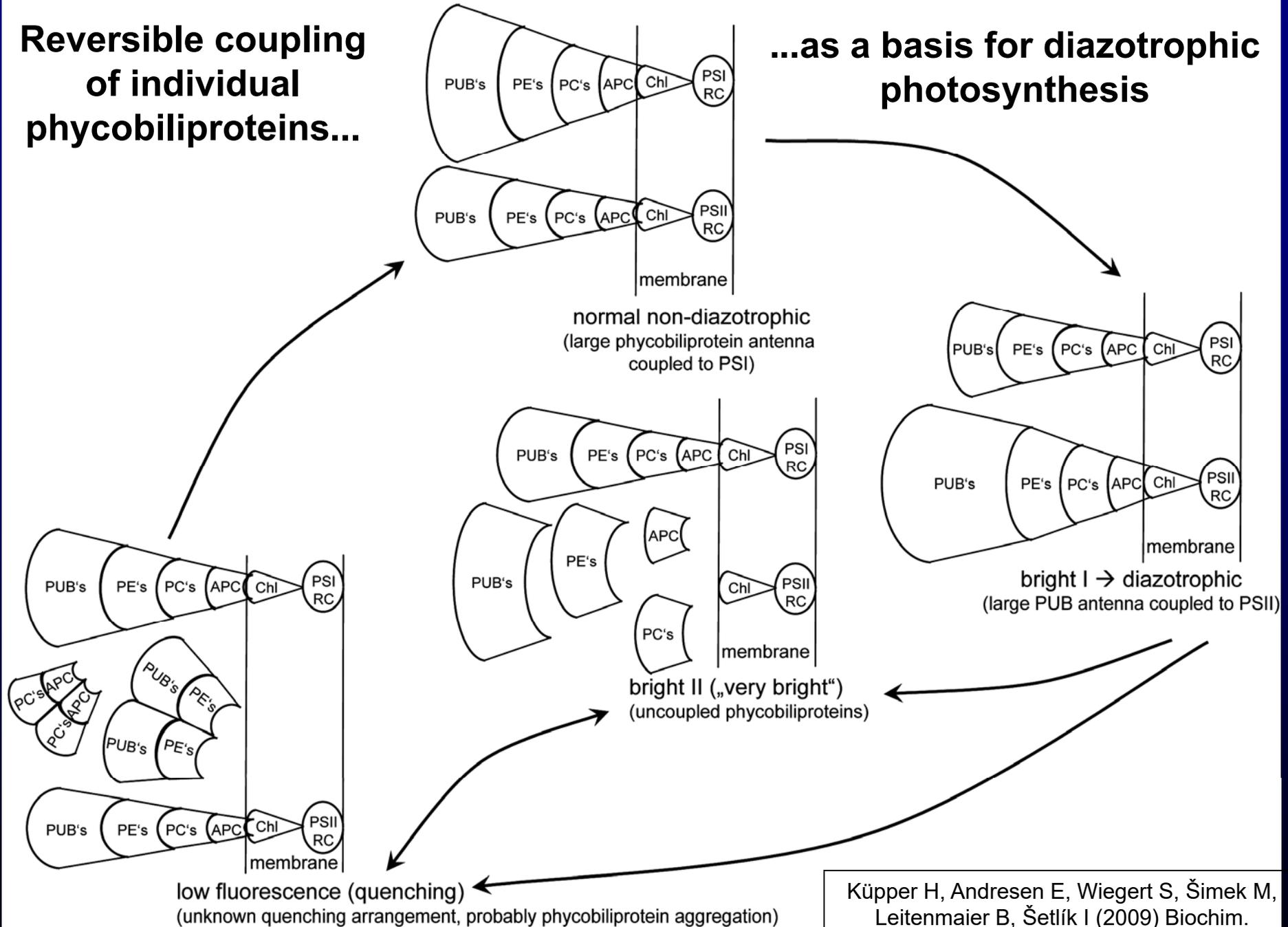
State 1

State 2



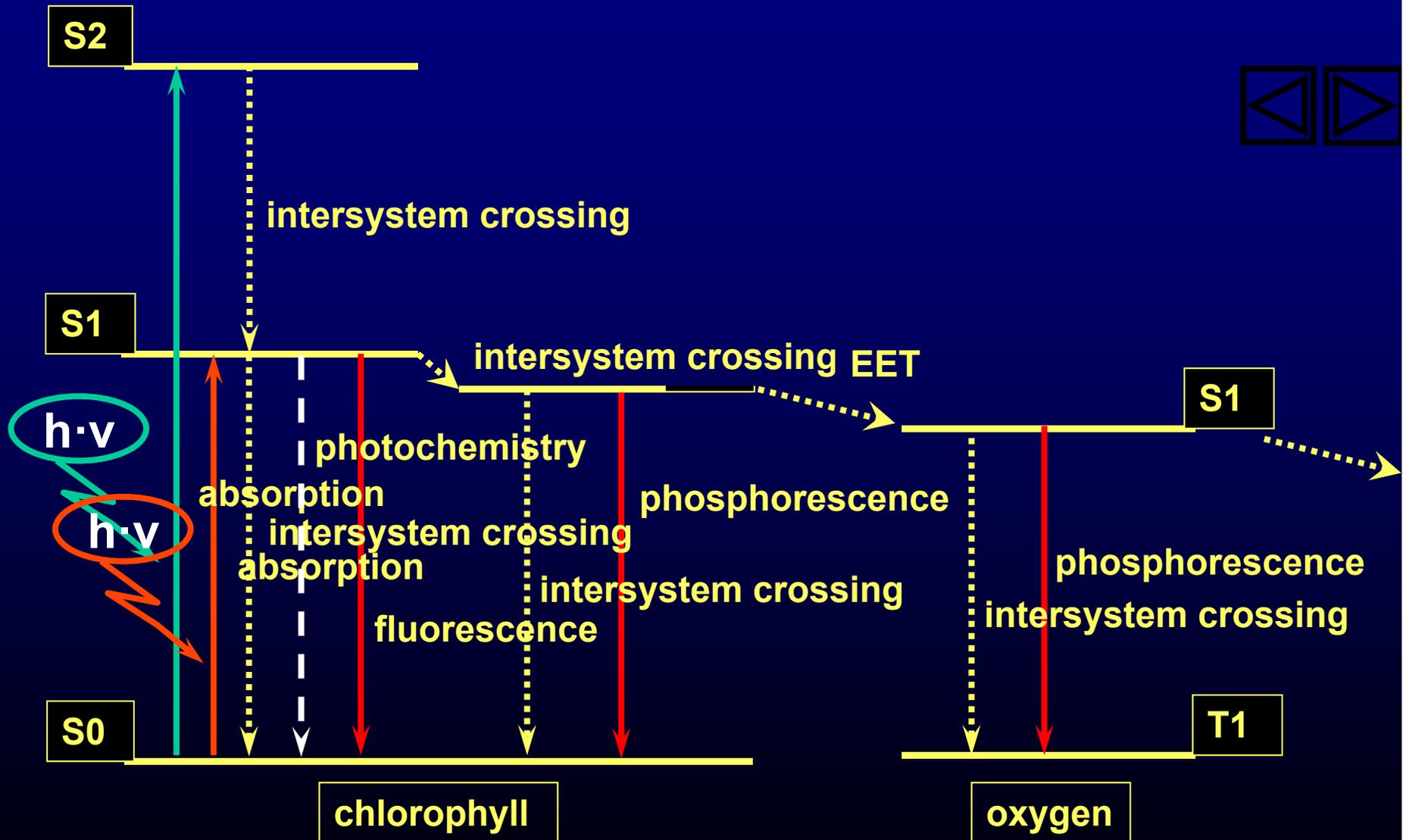
Reversible coupling of individual phycobiliproteins...

...as a basis for diazotrophic photosynthesis



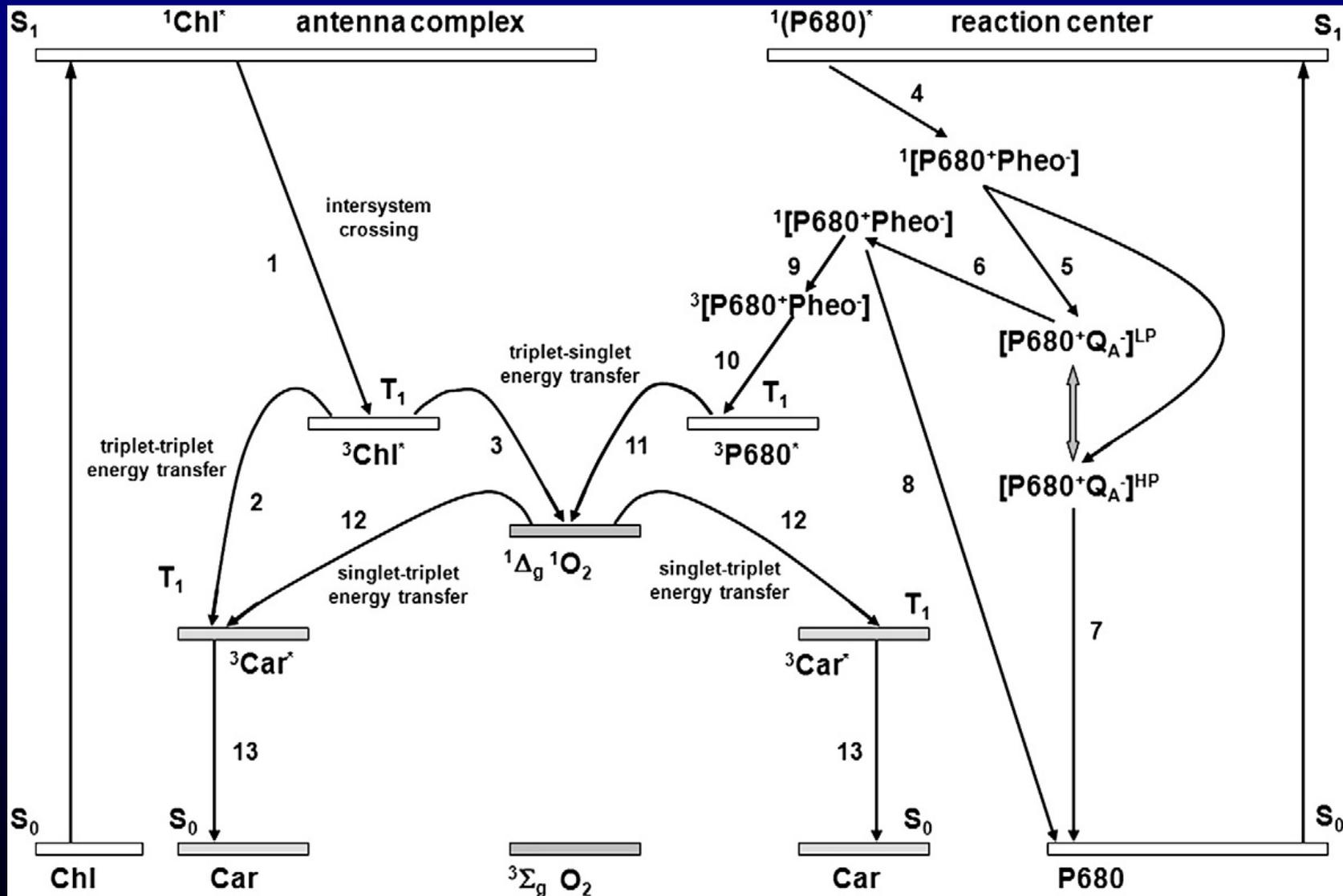
Küpper H, Andresen E, Wiegert S, Šimek M, Leitenmaier B, Šetlík I (2009) Biochim. Biophys. Acta (Bioenergetics) 1787, 155-167

Excitation energy transfer between chlorophyll derivatives and singlet oxygen



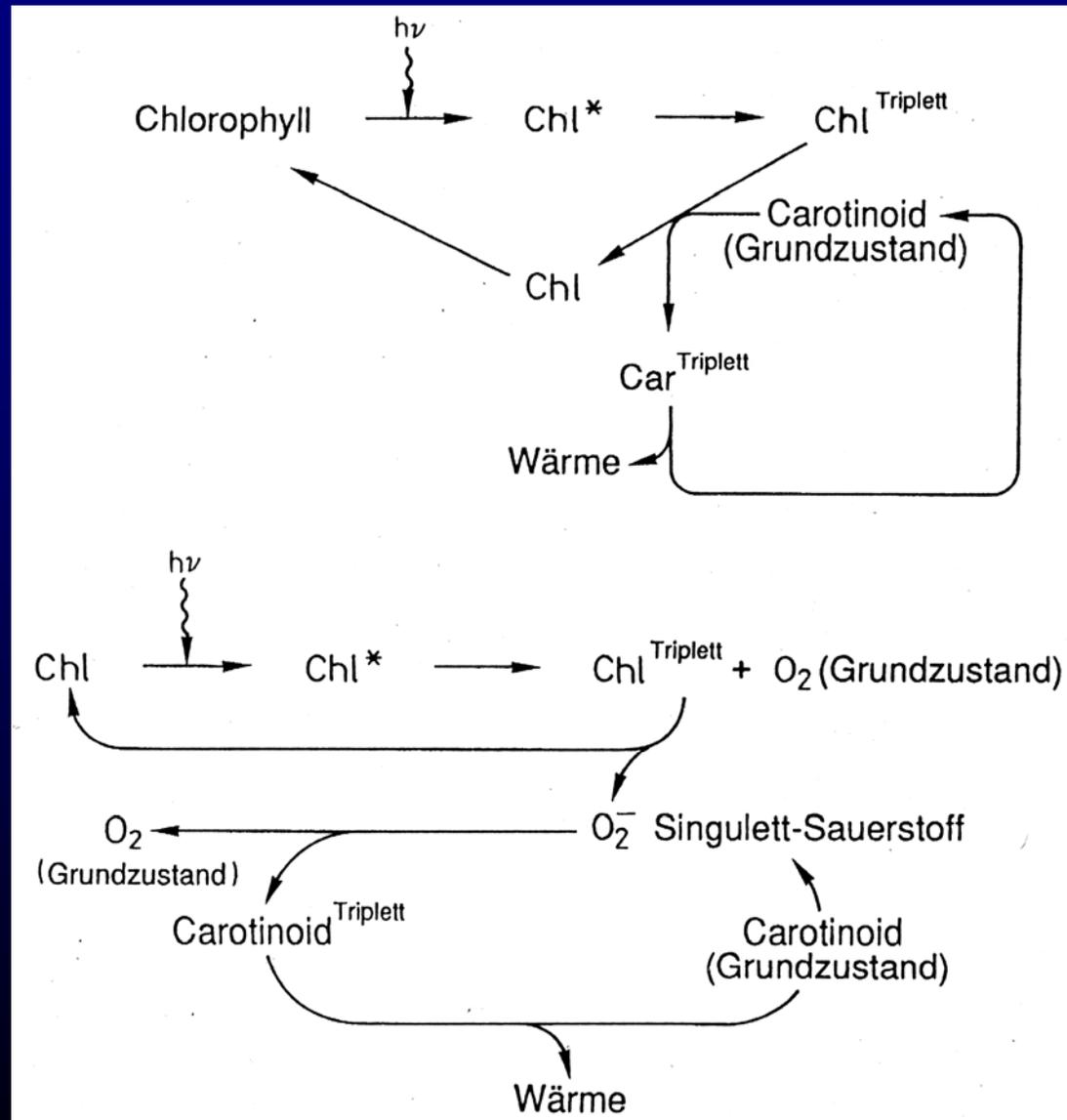
Photosynthesis related Proteins with metal centres

1. LHCII & PSII RC: generation & quenching of 1O_2



From: Pospisil P (2012) Biochimica et Biophysica Acta 1817, 218-31

Regulation of energy transfer (II): Mechanisms of protection by carotenoids against singlet oxygen

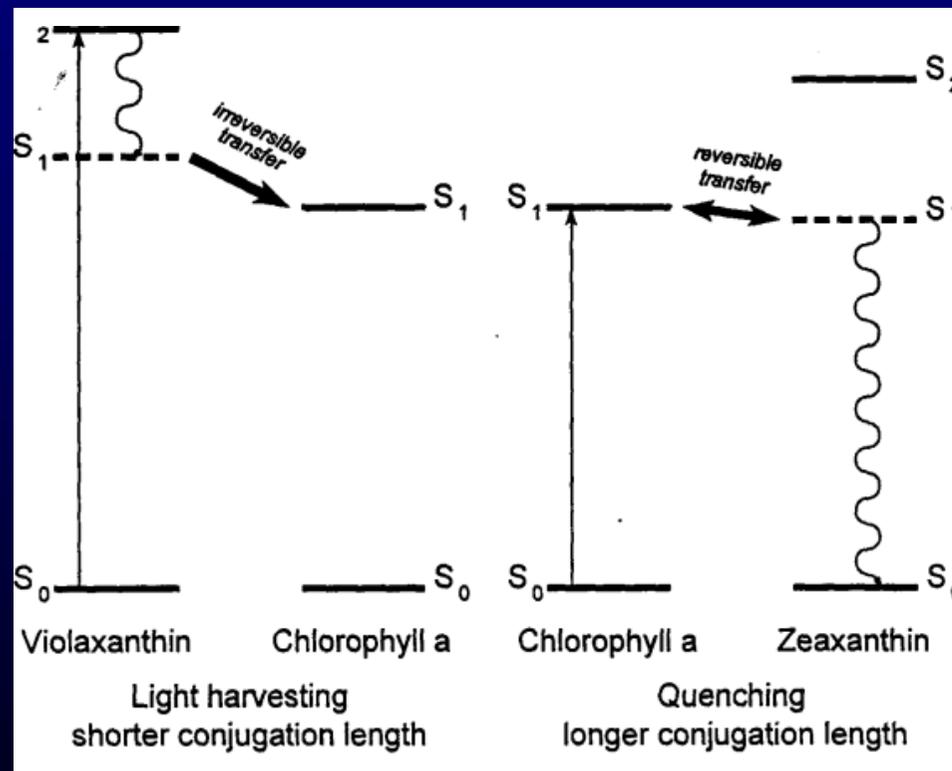


From: Lawlor DW (1990) Thieme, Stuttgart, 377S

Regulation of energy transfer: xanthophyll cycle

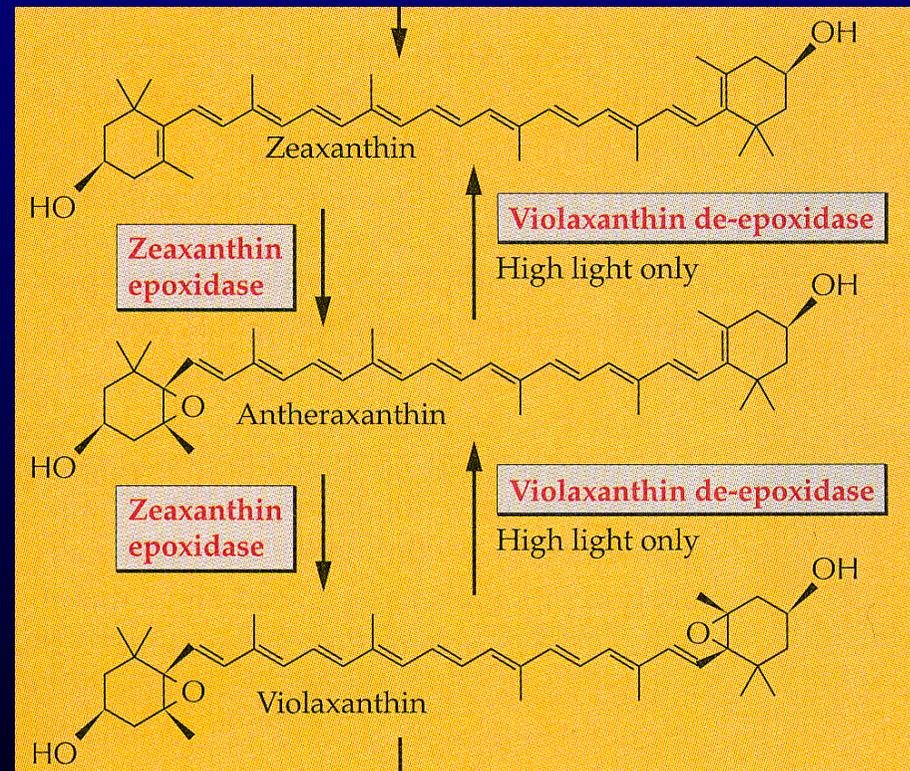
little light

much light

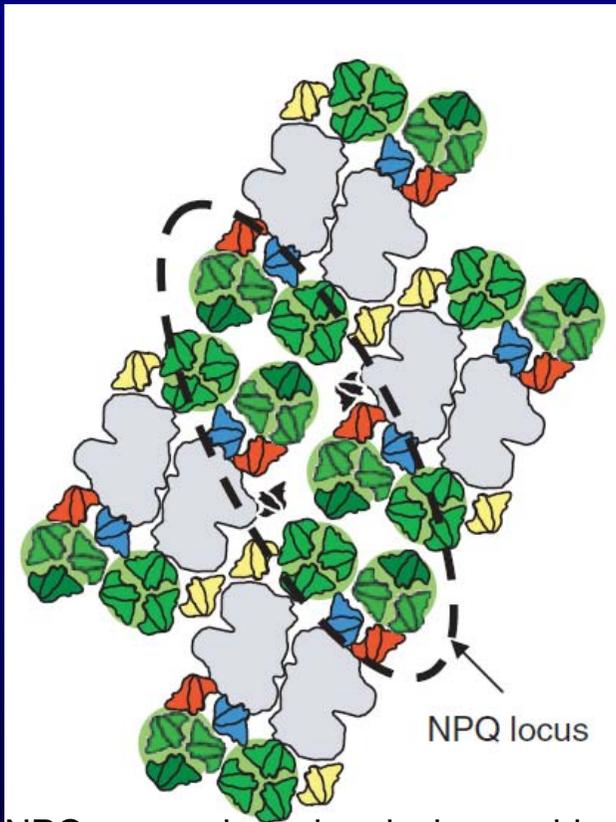


little light

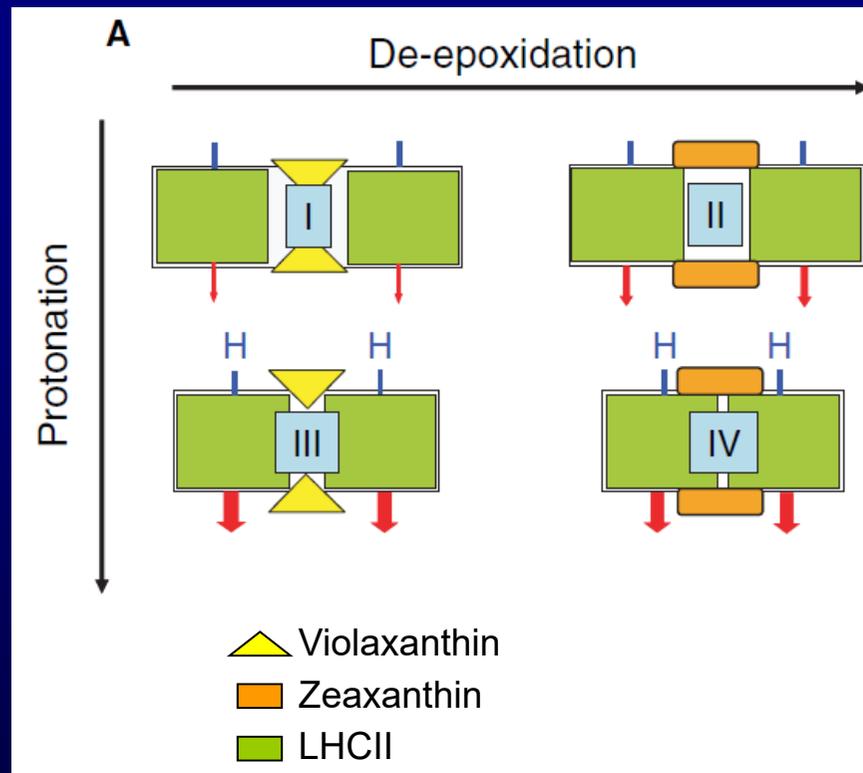
much light



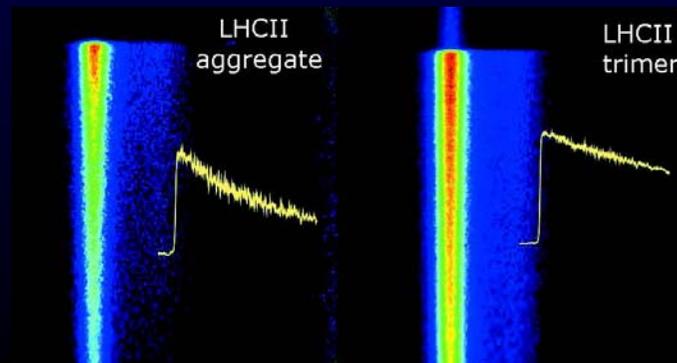
Fast adaptation to irradiance changes: combination of LHCII-aggregation with xanthophyll cycle



NPQ = non-photochemical quenching

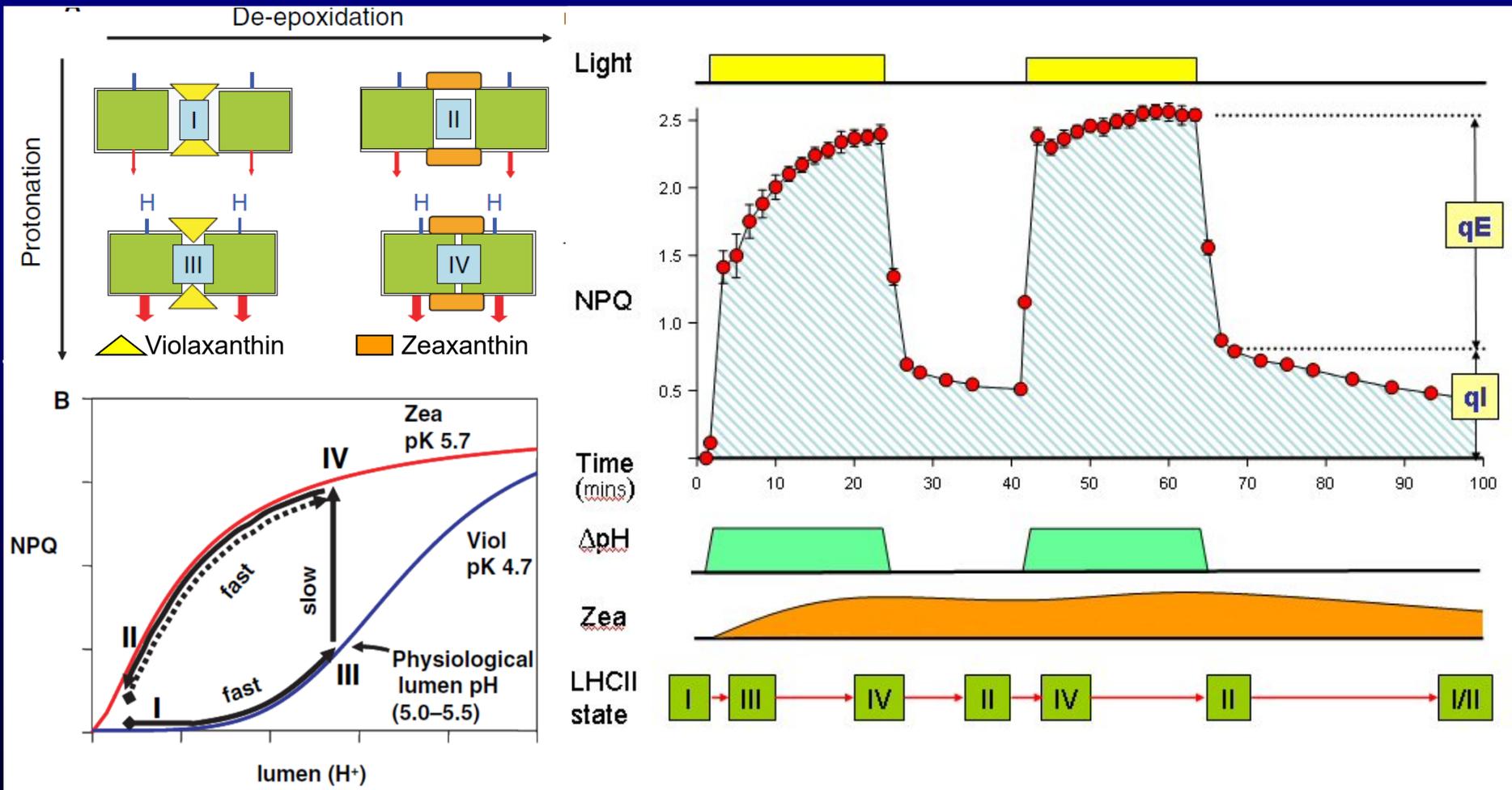


From: <http://photosynthesis.peterhorton.eu/research/ligtharvesting.aspx>
 Horton P, Johnson MP, Perez-Bueno ML, Kiss AZ, Ruban AV (2008) FEBS Journal 275, 1069-79



From:
http://www.laserlab.vu.nl/en/Research/research_projects/the_primary_processes_in_photosynthesis.asp

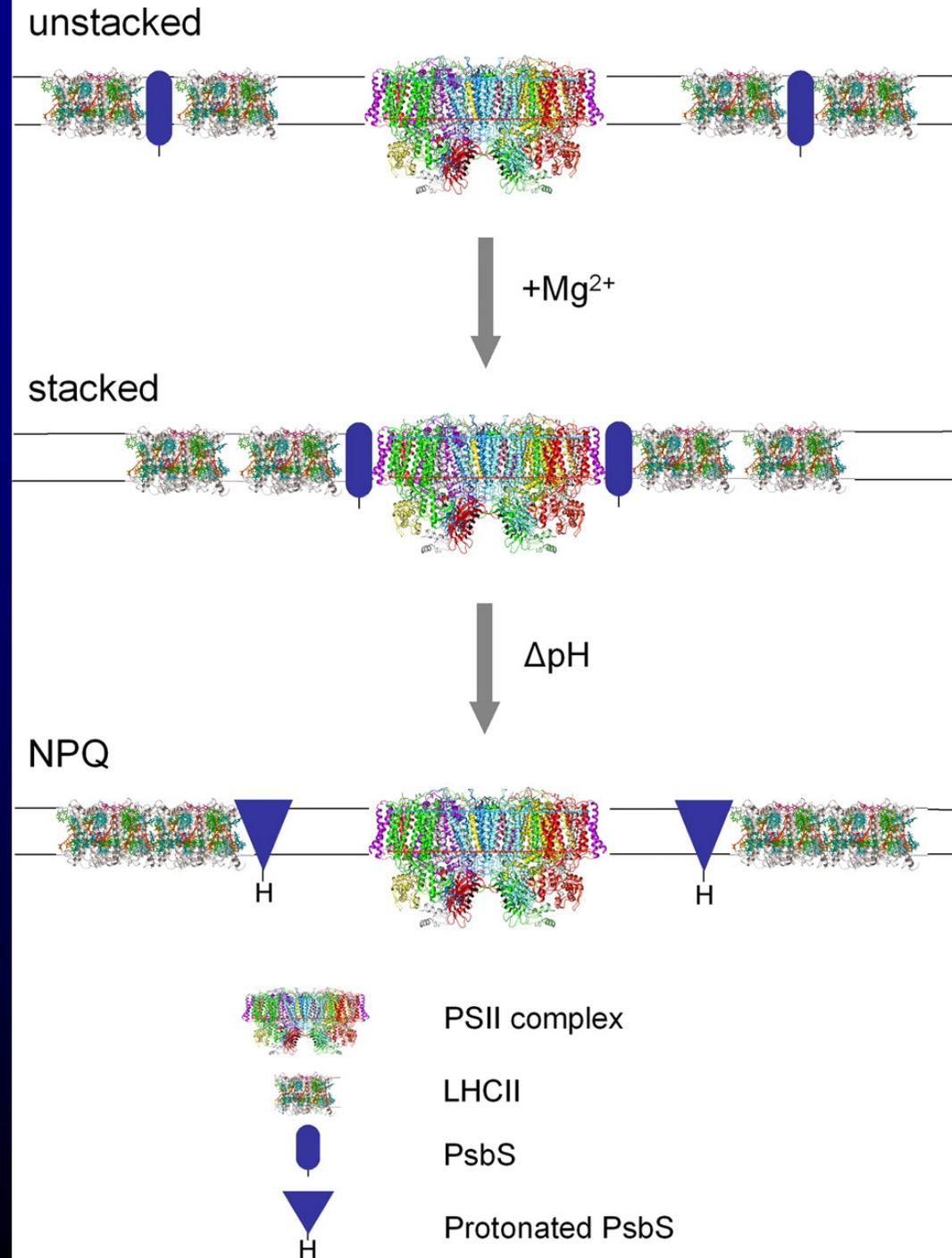
Fast adaptation to irradiance changes: combination of LHCII-aggregation with xanthophyll cycle



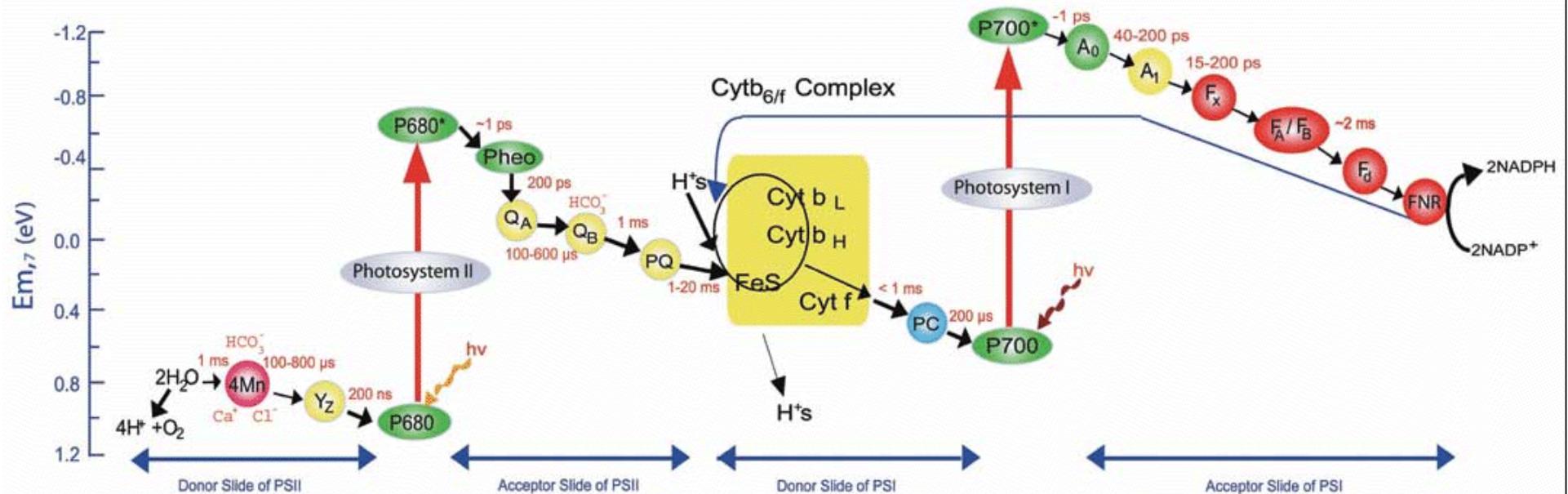
From: <http://photosynthesis.peterhorton.eu/research/ligtharvesting.aspx> (Horton lab web page)
 Horton P, Johnson MP, Perez-Bueno ML, Kiss AZ, Ruban AV (2008) FEBS Journal 275, 1069-79

PsbS modulation of the structure and function of the PSII antenna

- At relatively high but not inhibitory light, relatively many unstacked grana exist, where LHCII is not efficiently coupled to PSIIRC
- At low (limiting) light, enhanced grana stacking occurs, regulated via an increase of Mg^{2+} .
- At inhibitory high light, grana unstack again, and in addition protonation of PsbS leads to strong non-photochemical quenching of excitons



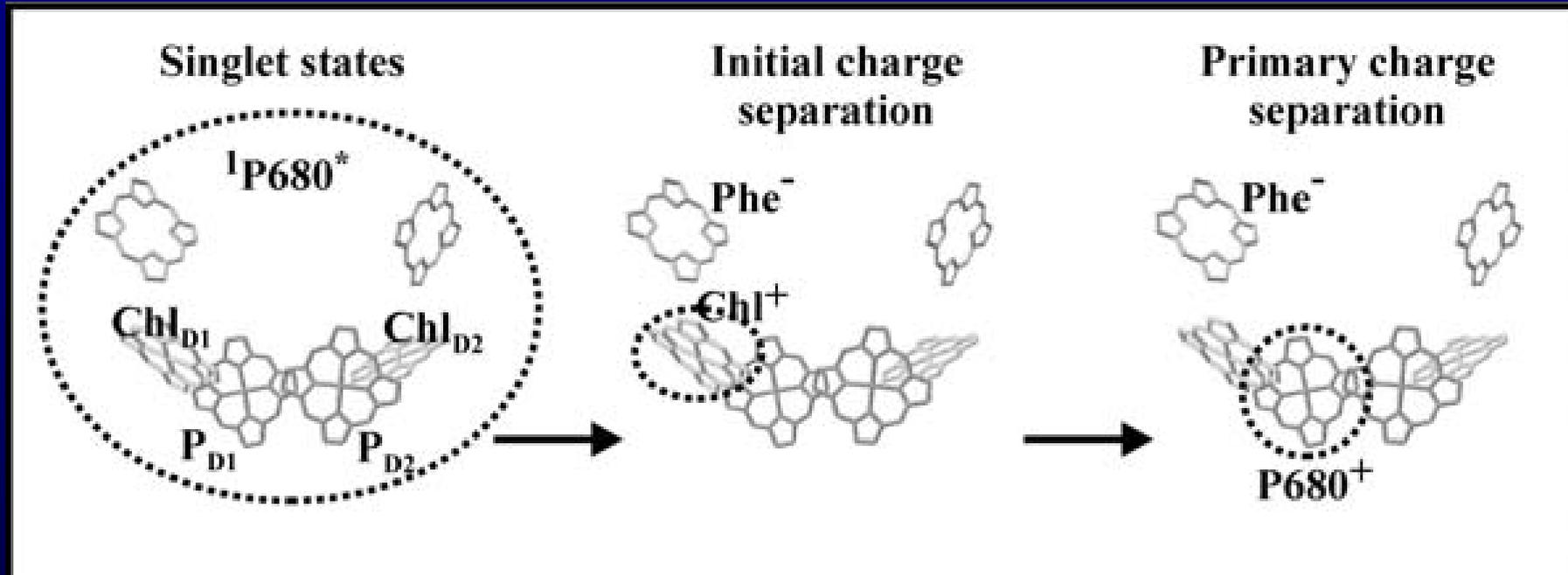
Overview of photosynthetic light reactions the „Z-scheme“



From: accessscience.com

Biophysical aspects of photosynthetic electron transport

A) Photosystem II reaction centre: special pair chlorophyll and pheophytins



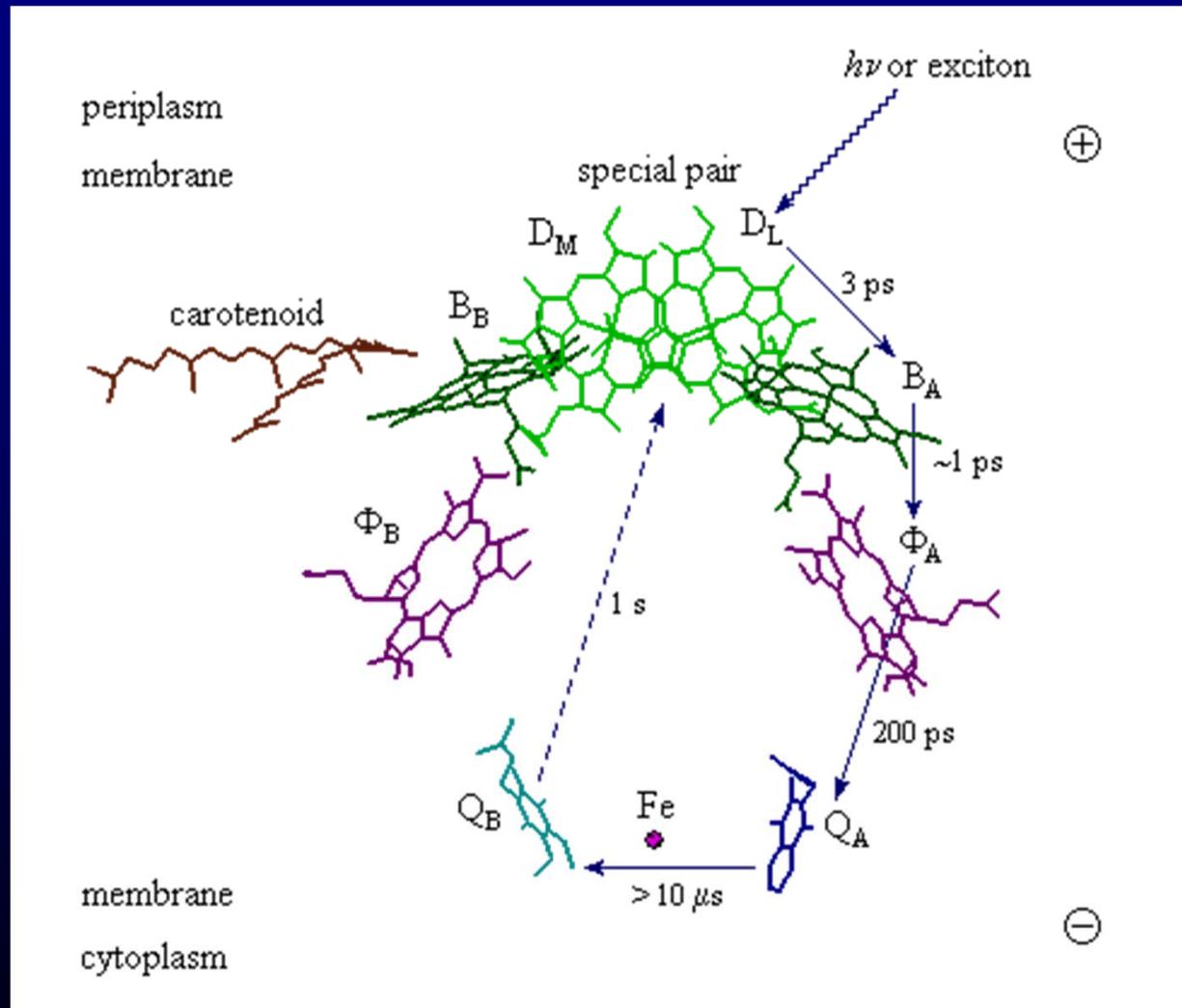
From: Barber J, 2003, QuartRevBiophys36, 71-89

Mechanism of charge separation

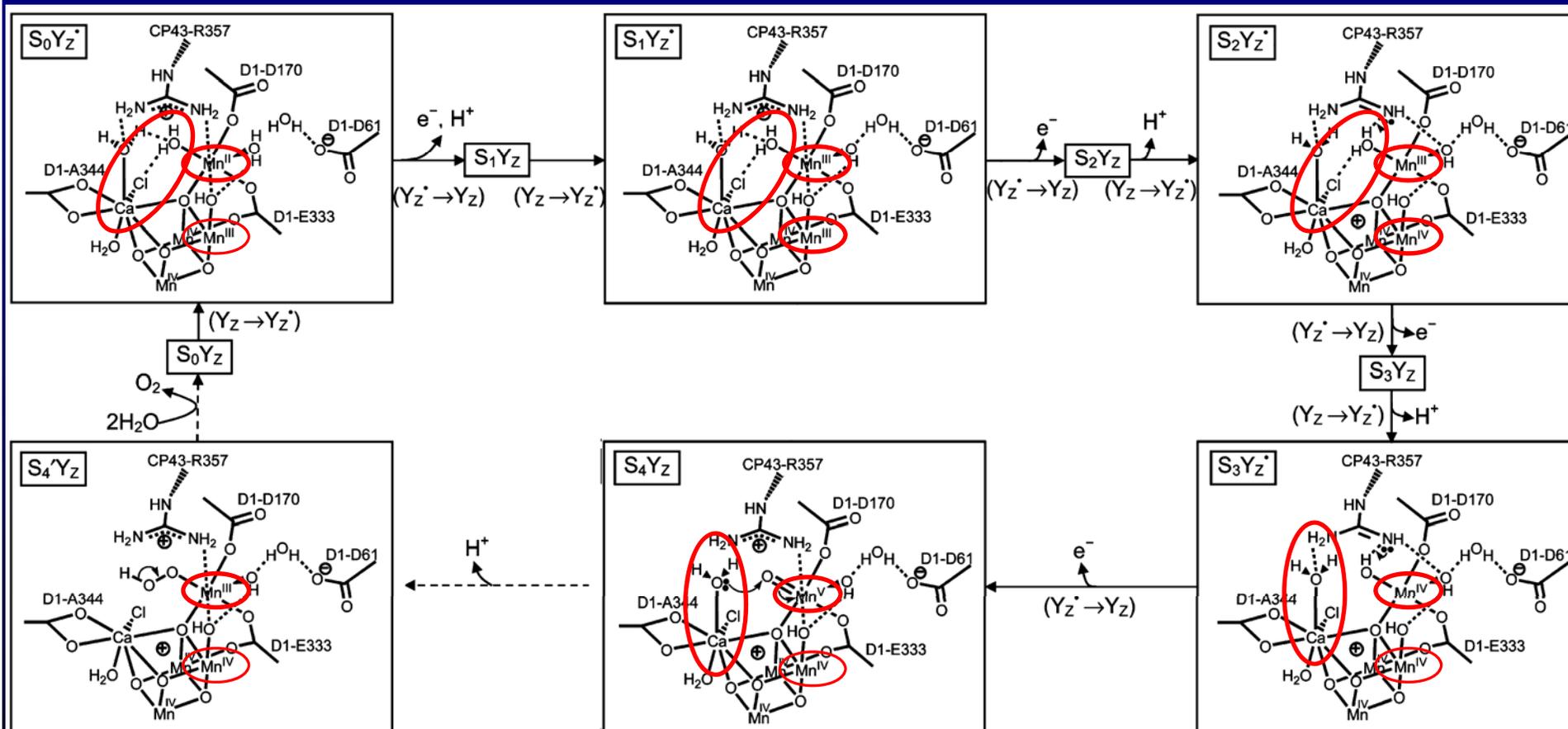
- 1. Special pair chlorophylls (=P680) accept excitons from antenna
- 2. Chl_{D1} transfers an electron to Pheo ("initial charge separation")
- 3. Within a few ps, the electron hole in Chl_{D1} is filled from P680 ($\rightarrow P680^+ / Pheo^-$) "primary charge separation"
- (according to other authors, the initial charge separation is in P680, and Chl_{D1} transfers the electron to Pheo, see next scheme...)

Biophysical aspects of photosynthetic electron transport

A) Photosystem II reaction centre: speeds of electron transfer



Water splitting complex of the photosystem II reaction centre proposed mechanism



From: McEvoy JP, Brudvig GW, 2006, Chemical Reviews 106, 4455-83

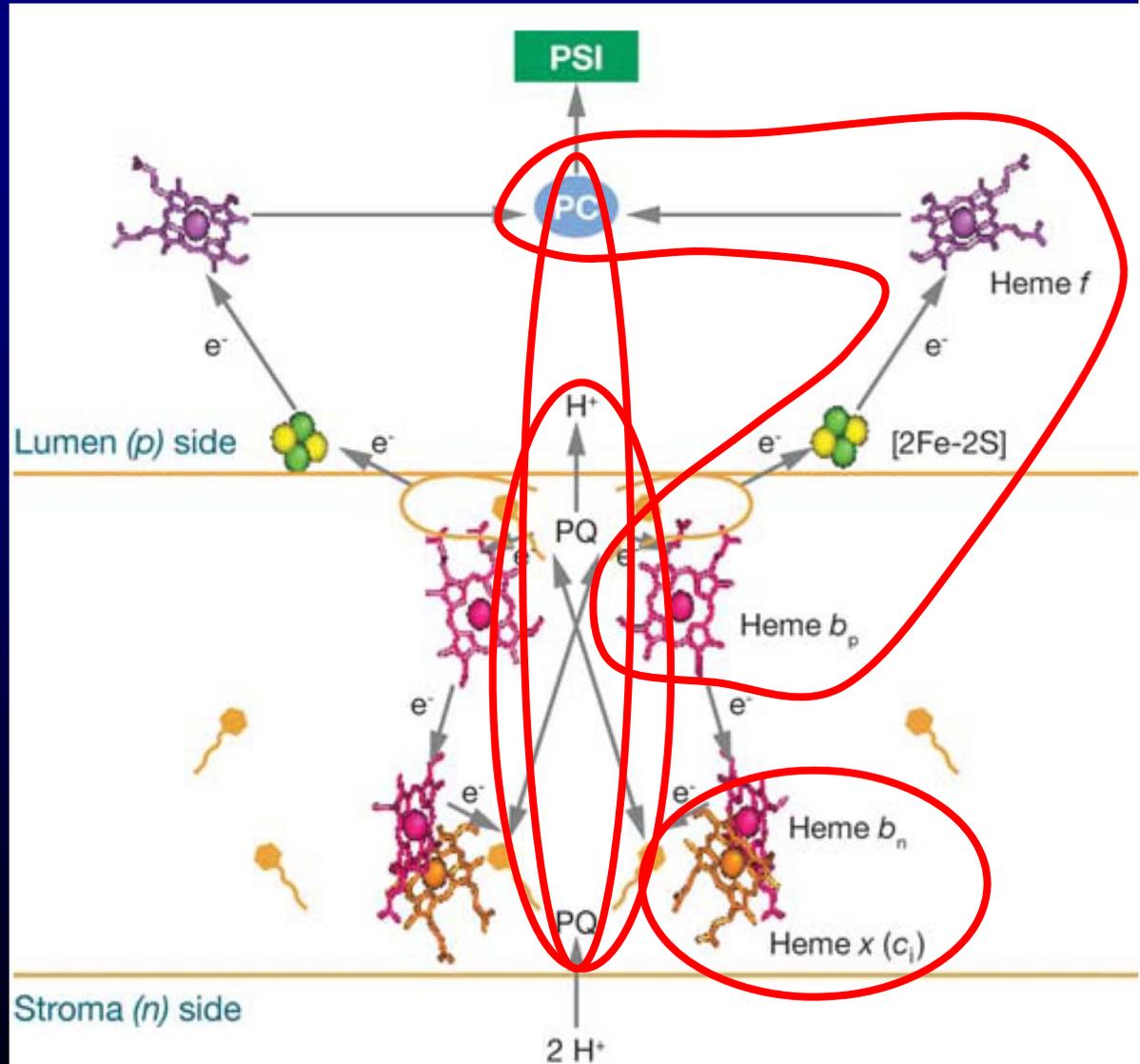
- 2 of the 4 Mn ions are redox-active ($3+/4+$), accepting electrons from water and transferring them to P680
- Ca^{2+} helps in binding the water

Biophysical aspects of photosynthetic electron transport

B) Cytochrome b_6f complex: mechanism

Functional characteristics

- transfers e^- from PQ to plastocyanin (PC),
- It uses the difference in potential between Q_B and PC for translocating a proton via 2x2 heme b groups and 2x1 heme x group
- Electrons are transferred from the heme b groups to PC via a “Rieske” $[2Fe_2S]$ -cluster and a heme f group
- Cyclic electron transport occurs via coupling of ferredoxin to heme x



Biophysical aspects of photosynthetic electron transport

C) Plastocyanin

Functional characteristics

- Oxidised (Cu^{2+}) plastocyanin accepts electron from Cyt_{b6f} complex,
- Reduced ($\rightarrow \text{Cu}^+$) plastocyanin diffuses to the PSIRC
- Plastocyanin releases the electron ($\text{Cu}^+ \rightarrow \text{Cu}^{2+}$)
- rigid protein structure facilitates fast red/ox-changes, but recent data show that copper binding still causes changes in structure (“induced rack” rather than “entatic state”)

From: Shibata N, Inoue T, Nagano C, Nishio N, Kohzuma T, Onodera K, Yoshizaki F, Sugimura Y, Kai Y, 1999, J Biol Chem. 274: 4225-30

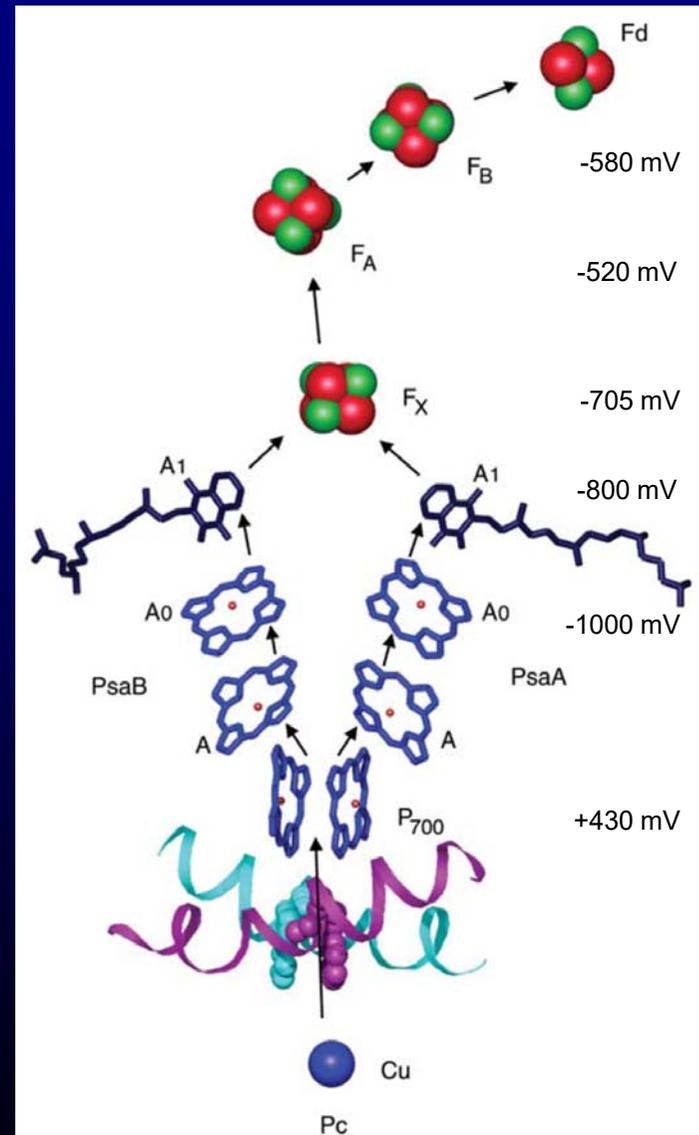


Biophysical aspects of photosynthetic electron transport

D) Photosystem I reaction centre

Functional characteristics:

- primary charge separation: special pair (=P700, Chl a / Chl a' heterodimer), releases e^- to A_0 via A (both Chl a)
- e^- transport via A1 (phylloquinone) and the [4Fe4S]-clusters F_x , F_A and F_B to the [4Fe4S]-cluster of ferredoxin
- P700 is re-reduced by plastocyanin



**All slides of my lectures can be downloaded
from my workgroup homepage**

Biology Centre CAS → Institute of Plant Molecular Biology → Departments
→ Department of Plant Biophysics and Biochemistry,
or directly

http://webserver.umbr.cas.cz/~kupper/AG_Kuepper_Homepage.html